Synthesis of fused 1,2,5-triazepine-1,5-diones and some N^2 - and N^3 -substituted derivatives: potential conformational mimetics for *cis*-peptidyl prolinamides¹

Morag M. Lenman, Arwel Lewis and David Gani*

School of Chemistry and Centre for Biomolecular Sciences, The Purdie Building, The University, St Andrews, Fife, UK KY16 9ST

The synthesis of a new fused 1,2,5-triazepine-1,5-dione heterocycle, which is expected to mimic structural features of *cis*-peptidyl prolinamides, is described. The required parent heterocycle, corresponding to *cis*-glycyl-(2.*S*)-prolinamide, has been prepared in good yield by the cyclisation of *N*-(2-bromoacetylprolyl)-hydrazine which is itself generated *in situ* from the bromoacetyl proline methyl ester. Analogues corresponding to *cis*-(2*R*)-alanyl- and *cis*-(2*S*)-prolinamide have been similarly prepared from the appropriate *N*-(2-bromopropionyl)proline methyl esters and hydrazine hydrate where the cyclisation step, involving the displacement of bromide, has been shown to occur with inversion of configuration at C-2 of the propionyl moiety. Acylation at the N-3 position of the triazepine is equivalent to N-terminal acylation of the residue preceding the proline residue in *cis*-aminoacyl prolinamides. This has been achieved without incident using standard peptide coupling procedures. Extension at the 'C-terminal' has been achieved by preparing elaborated hydrazine precursors which are reacted with suitably activated esters of *N*-a-halogenoacylprolines, prior to cyclisation, to give the required fused triazepine dione. Thus it is possible to prepare constrained *cis*-peptidyl prolyl peptide mimetics of defined stereochemistry based upon this new triazepine dione in which all of the non-proline residues can be varied.

Introduction

Proline is unique amongst proteogenic amino acids in possessing a secondary amino group. Acyl prolines possess no amide hydrogen atoms and, therefore, the energies of the *cis*- and *trans*-isomeric forms are similar.² In nature, the Xaa–Pro peptide bond can exist in both stable conformations, structures **1** and **2**, and both forms occur in proteins, as in ribonuclease,³ and in bioactive peptides, such as bradykinin.⁴ The interconversions of the *cis*- and *trans*-forms of small peptides are quite slow in water⁵ ($k = 10^{-1}-10^{-3}$ s⁻¹ at 30 °C) and peptidylprolyl *cis*-*trans*-isomerases (rotamases), for example cyclophilin⁶ and FK506 binding protein,⁷ exist to speed up the rates of isomerisation.⁸ In recent years there has been immense interest in the enzyme activities and it is now believed that the slow uncatalysed isomerisation rate is important and may play a role in controlled proteolysis.



Given that some proline-containing bioactive peptides including angiotensin and thyroliberin might bind to their receptors with the Pro residue fixed in the *cis*-form⁹ it has been of interest to synthesise constrained cis-peptidyl prolinamides which might serve as high affinity ligands. In all examples of previously reported constrained *cis*-peptidyl proline analogues. the constraining modification had caused a significant increase in the steric size of the system, for example, the C^{α} - C^{α} bridged type VI reverse-turn Gly-Pro mimetics recently described by Gramberg and Robinson (compound 3)¹⁰ and by McEnaney and Curran (compound 4).11 We wished to introduce the smallest modification possible in order to retain the maximum potential for biological activity in the finished peptides and/or structural motifs of proteins. Therefore, we opted to link the N^{α} -atom of the residue preceding Pro to the N^{α} -atom of the residue following Pro. Such a modification differs from the parent system 2 by just two H-atoms, but fixes the stereochemistry of the acyl proline amide bond in its cis- or (E)configuration through the formation of a novel [d]-fused 1,2,5triazepine-1,5-dione system 5. This compound mimics a cispeptidyl prolinamide, and as such might be used as a reverseturn mimetic in the design of biologically active molecules that need to emulate features of the common β -turn motifs (e.g. type I, II, IV). β-Turns are not only important structural features in protein secondary structure but are implicated in the recep-tor bound conformations of many bioactive peptides.^{4,12,13} Recently, work on the synthesis of systems designed to mimic the various types of β -turns include indole derivatives,¹⁴ α alkylated aspartic and glutamic acids,¹⁵ azanorbornane deriv-atives¹⁶ and acetylenes.^{17,18}

Results and discussion

To synthesise the simplest [*d*]-fused 1,2,5-triazepine-1,5-dione **6**, (2.*S*)-proline methyl ester **7** was treated with bromoacetic acid **8** activated as its mixed anhydride to give the required bromoacetamide **9**, Scheme 1. Treatment with hydrazine hydrate in ethanol was expected to give an equilibrium mixture of predominately the *trans*- α -bromoacetyl proline hydrazide **10** and



Scheme 1 Reagents and conditions: i, NMM, isobutyl chloroformate, THF-DMF, -40 °C; ii, H₂NNH₂·H₂O, EtOH, reflux

the *cis*-rotomer which would cyclise through the nucleophilic displacement of bromide ion in the *cis*-rotomer by the non-acylated hydrazine N-atom, to give the required *cis*-glycyl (2.*S*)-prolinamide analogue **6**. In the event, after refluxing for 1.5 h a compound was isolated which gave the expected mass data and showed the existence of two major conformational isomers in its ¹H and ¹³C NMR spectra in [²H₆]DMSO. The compound failed to form a hydrazone adduct upon treatment with benzaldehyde and together these results indicated that the required 7-membered triazepine rather than a 6-membered piperazine ring had been formed.

Each of the diastereomeric (2S)- and (2R)-alanyl-(2S)proline ester homologues, 9a and 9b, were prepared in a similar manner, starting from the appropriate chiral 2bromopropionic acids, which were themselves prepared via the diazotisation-bromination of (2S)- and (2R)-alanine.¹ Each diastereomeric ester 9a and 9b was obtained as a crystalline solid in moderate yield but examination of the NMR spectra of the crude reaction mixtures for each of the coupling reactions indicated that some epimerisation had occurred. In subsequent reactions the epimerisation was largely prevented by lowering the temperature of both the acyl group activation and N-acylation reaction from -15 to -40 °C. For the compound **9b**, that derived from (2R)-2-bromopropionic acid, the absolute stereochemistry at C-2 of the 2-bromopropionyl moiety was verified by X-ray crystallography using the (2S)configuration of the proline moiety as a stereochemical reference.1

Treatment of each of the individual esters 9a and 9b with hydrazine hydrate in refluxing ethanol gave a new product almost immediately, as judged by TLC and by NMR spectroscopy, but in each case, reaction to give a second new product was not complete until 16 h had elapsed, Scheme 1. Thus, it appeared that for each of the homologues 10a and 10b the cyclisation step was significantly slower than for the nonmethylated derivative 10. These observations are almost certainly explained by (i) the low equilibrium concentration of the cis-isomers of 10a and 10b which are required for the cyclisation, due to the steric effects of the extra methyl groups (compared to hydrazide 10), and (ii) the fact that the cyclisation itself occurs via S_N2 attack on a secondary a-bromoacylamide instead of a primary a-bromoacylamide, as in the case for hydrazide 10. Nevertheless, after prolonged reaction, 16 h in refluxing ethanol, each of the triazepines 6a and 6b were obtained in excellent yield (>90%) and each compound displayed the expected spectral and analytical properties.

In order to confirm the expectation that the cyclisations had proceeded with inversion of configuration at C-2 of the bromopropionyl moiety, the fused triazepines **6a** and **6b** were each subjected to dissolving metal reduction to cleave the N^2-N^3 bond and give the alanyl prolinamides **11a** and **11b**. Each of these compounds underwent an extremely rapid spontaneous cyclisation to give the fused dioxopiperazines **12a** and **12b** through the displacement of ammonia from the Pro carboxamide moiety, Scheme 2, which we now wished to compare



Scheme 2 $\,$ Reagents and conditions: i, Na–NH_{3(liq)}, $-68~^\circ C;$ ii, H2, Pd–C, MeOH, 22 $^\circ C$

with authentic materials. Each of the diastereomeric dioxopiperazines 12a and 12b were therefore also prepared through the catalytic hydrogenolysis of the terminal benzyloxycarbonyl (Z) protecting group of samples of the N-Z-alanyl-(2S)-proline esters¹⁹ 13a and 13b. The same compounds were also prepared from samples of the corresponding N-Z-alanyl-(2S)-prolinamides 14a and 14b, where in each case the amine product rapidly cyclised to the lactam, Scheme 2. The dipeptide esters and amides were prepared using standard solution phase coupling and protection strategies and displayed the expected spectral and analytical data. Comparison of the ¹H and ¹³C NMR spectra of these dioxopiperazines indicated that the cyclisation of the hydrazides 10a and 10b to form the triazepines 6a and 6b (see Scheme 1) had occurred with inversion at the secondary carbon centre. Therefore, it was possible to synthesise cisaminoacyl prolinamide mimetics of defined stereochemistry.

The reason that we had synthesised the protected dipeptide amides **14**, **14a** and **14b** was because the dissolving metal reduction of the triazepinediones **6**, **6a** and **6b** and their subsequent lactamisation reactions had been so facile. Essentially, no uncyclised material remained after work-up or after low temperature (-50 °C) NMR spectroscopic analysis of the reaction quenched with [²H₄]methanol (after the blue colour due to solvated electrons had disappeared). Since cleavage of the hydrazine moiety under such conditions could generate highly nucleophilic metal amides or N-centred radicals derived from N-3, we wished to assess how rapidly the amines **11**, **11a** and **11b** derived from the protected dipeptide amides **14**, **14a** and **14b**

would cyclise. In the event, small quantities of uncyclised amine **11**, **11a** and **11b** could be detected as judged by ¹H NMR spectroscopy, after partial but rapid catalytic hydrogenolysis under neutral conditions at room temperature (20-25 °C). Thus, this finding appears to suggest that much more of the non-cyclised material 11, 11a and 11b should have been detected in quenched dissolving metal reductions of compounds 6, 6a and 6b, if the cyclisation involved the simple displacement of ammonia by a neutral amine nucleophile. A further allowance for the differences in the cyclisation rates must take into account the fact that at low temperature, all of the nascent amino amides derived from compounds **6**, **6a** and **6b** is in the *cis*-rotomer form and cannot isomerise to the *trans*-form. Thus, we are currently exploring less 'basic' conditions for generating the *cis*-forms of the amino amides 11, 11a and 11b in order to evaluate each of the factors involved in determining the cyclisation rate.

The next objective in the synthesis of *cis*-peptidyl prolinamide mimetics was to extend both the N- and C-termini of the dipeptide mimetic, corresponding to N-2 and N-3 of the triazepine, to emulate longer and more elaborate polypeptides.

In order to achieve these ends, it was necessary to determine the regioselectivity for the reaction of unsymmetrical hydrazines with the two electrophilic centres present in the bromoacylprolinate esters **9**, **9a** and **9b**. Accordingly, each of the α -bromoacyl-(2.*S*)-proline methyl esters **9**, **9a** and **9b** were treated with methylhydrazine. Unfortunately, but as expected, the ester methoxy group was displaced exclusively by the less hindered primary amino group and subsequent cyclisation occurred with inversion of configuration to give the N^8 -methyl-1,2,5-triazepine diones **15**, **15a** and **15b**. Since these compounds



were resistant to both dissolving metal and catalytic reduction, the structures of compounds **15** and **15a** were verified by X-ray crystallographic analysis using the known stereochemistry at C-2 of proline for reference.¹ The outcome of these reactions indicated that we would need to introduce alkyl groups at the N-2 position using a more elaborate approach, see below.

In order to assess the reactivity of the N-3 position of the triazepine which is equivalent to the N-terminal of a *cis*dipeptide, compound **6a** was treated with acetic anhydride in pyridine. After 2 days the N^{s} -acetyl-1,2,5-triazepine-1,5-dione **16** was isolated in moderate yield. Similarly the reaction of compound **17a** (see below) with *N*-Z-(2.*S*)-phenylalanine activated using isobutyl chloroformate and *N*-methylmorpholine gave the corresponding N^{s} -phenylalanyl derivative **17c**, indicating that extension at the N-terminal of the peptide mimetic was possible.

Given the preferred regioselectivity for the reaction of

methylhydrazine with compounds **9**, **9a** and **9b**, it was evident that extension at the proline carboxamide (substitution at the N^3 -position of the triazepine-1,5-dione system) would require the synthesis of unsymmetrical alkylhydrazines protected on the primary amino group. Accordingly, N^2 -tert-butoxy-carbonyl- N^4 -methylhydrazine **18** was prepared from methylhydrazine in three steps. Reaction of this with the mixed anhydride of chloroacetyl-(2*S*)-proline **19**, Scheme 3, gave the



Scheme 3 Reagents and conditions: i, ClCH₂COCl, NaHCO₃, EtOAc, 78 °C, 50 min, 74%; ii, NMM, isobutyl chloroformate, THF, -10 °C, 12 h, 73%; iii, HCl, MeOH, 22 °C; 15 min; then NaOH, H₂O, 5 min, 79%

required proline hydrazide **20** in 73% yield. Our reasons for using the chloroacetyl rather than bromoacetyl derivatives are given below. Removal of the *tert*-butoxycarbonyl (Boc) protecting group with hydrogen chloride gave the amine salt which, upon treatment with sodium hydroxide, cyclised to give the desired N^{\sharp} -methyl-1,2,5-triazepine-1,5-dione **17**.

The synthesis of the 4-methyl- N^2 -methyl-1,2,5-triazepine-1,5-diones 17a and 17b proved to be more difficult due to unwanted reactions at the second chiral centre (C-2 of the propionyl moiety). The carboxylic acid precursors 23a and 23b were obtained through the saponification of the methyl esters 22a and 22b respectively, using 1 mol dm⁻³ aqueous sodium hydroxide, Scheme 4. It is worth noting that hydrolysis of the corresponding bromo dipeptide esters (see above) leads to products that contain two sets of signals in the ¹H and ¹³C NMR spectra. These were later shown to be due to the presence of both diastereomers, and therefore it was reasoned that the C-2 brominated centre had undergone base-catalysed enolisation which had resulted in epimerisation. The 2-chloropropionyl analogues 22a and 22b were found to be much more stable and could be saponified without epimerisation, see below.

The (2.5)- and (2*R*)-2-chloropropionic acids, **21a** and **21b** were prepared according to the method of Fu *et al.*²⁰ and were coupled with proline methyl ester using mixed anhydride methodology as for the bromide analogues. The methyl esters **22a** and **22b** were obtained in 71 and 83% yield respectively and base catalysed hydrolysis then gave the corresponding acids, **23a** and **23b**. The ¹H and ¹³C NMR spectra showed only one set of signals for each compound, indicating that no epimerisation had taken place. Reaction of each of the free acids, **23a** and **23b**, activated as their respective mixed anhydrides, with the methylhydrazine derivative **18** (see Scheme 4) gave the prolyl hydrazides **24a** and **24b** in 69 and 74% yield respectively after



Scheme 4 Reagents and conditions: i, NMM, isobutyl chloroformate, THF, -15 °C, 3 h, 83%; ii, 1 mol dm⁻³ NaOH, MeOH-H₂O, room temp., 1 h, 84%; iii, NMM, isobutyl chloroformate, THF, -15 °C, 12 h, 74%; iv, HCl, MeOH, 22 °C; 20 min, 100%; v, NMM, 22 °C; 5 min, 64%

purification by column chromatography. Each compound showed the expected spectroscopic and analytical data. Removal of the Boc groups gave the hygroscopic salts, **25a** and **25b**, which were cyclised in the presence of 2 equiv. of *N*-methylmorpholine (NMM) to give the N^2 -methyltriazepines **17a** and **17b**. These bicyclic compounds displayed appropriately different spectroscopic characteristics to the isomeric compounds **15a** and **15b** in accordance with the expected structures.

The N^2 -phenyl derivative **26** was prepared in four steps in an overall yield of 23% in an analogous manner to that for compound **17** starting from (2*S*)-proline and phenylhydrazine, Scheme 5. The relatively difficult coupling reaction for



Scheme 5 Reagents and conditions: i, Boc_2O , Et_2O , 22 °C, 67 h, 86%; ii, SOCl₂, pyridine, DMAP, CH_2Cl_2 , 22 °C, 2 days, 62%; iii, HCl, EtOAc, 22 °C; 30 min; then NaOH, H_2O , 30 min, 68%

chloroacetyl-(2.*S*)-proline **19** and hydrazide **27** was achieved by activation of the carboxylic acid as an acyl pyridinium salt. This intermediate was then used to acylate the phenylated N-atom of the hydrazide in good yield (62%), see Scheme 5.

Removal of the Boc protecting group using hydrogen chloride followed by treatment with aqueous sodium hydroxide gave the required N^2 -substituted bicycle **26**.

Thus, it was possible to prepare fused 1,2,5-triazepine diones with an alkyl or aryl group present at N-2. Since the ultimate objective was to introduce peptide extensions at the C-terminal end of the *cis*-peptidyl proline mimetic, we opted to prepare a structural analogue for a functional biological system, the *cis*glycyl prolyl phenylalanine methyl ester mimetic **29**, which is



 $\begin{array}{rrr} \mathbf{29} & \mathbf{R} = & \mathbf{H} \\ \mathbf{30} & \mathbf{R} = & \mathbf{ZAla} \end{array}$

part of the sequence of the self-cleaving polypeptide from the foot and mouth disease virus 2A region.^{21,22} Accordingly (2S)-phenylalanine 31 was treated with aqueous potassium cyanate to give the hydantoic acid 32, Scheme 6. Upon treatment with aqueous potassium hypochlorite this material underwent a Hofmann type rearrangement²³ to give (2.S)-Naminophenylalanine 33. Benzyloxycarbonyl protection of the primary amino group, followed by esterification, using thionyl chloride and methanol, gave the methyl ester hydrazide 35 in 60% yield from the hydrazino amino acid. Reaction of the hindered secondary amino group of this hydrazide moiety with the activated ester of N-chloroacetyl-(2S)-proline 19 using a variety of coupling reagents including isobutyl chloroformate, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI), 2-(1H-benzotriazole-1-yl)-1,1,3,3-tetramethyluronium tetrafluoroborate (TBTU), benzotriazole-1-yloxytris(pyrrolidino)phosphonium hexafluorophosphate (pyBOP) and diphenylphosphoryl azide (DPPA) all met with failure. However, when the preformed acyl pyridinium salt of 19 (prepared using thionyl chloride in the presence of pyridine) was treated with the free base of hydrazide 35 the reaction proceeded to give the required substituted peptidic ester 36. Under optimised conditions a yield of 53% was obtained and the compound displayed the expected properties. The removal of the benzyloxycarbonyl group from compound 36 was achieved using



Scheme 6 Reagents and conditions: i, KOCN, H_2O , 60 °C, 4 h; then HCl, H_2O , 86%; ii, KOCl, KOH, H_2O , 80 °C, 90 min; then HCl, H_2O , 36%; iii, benzyl chloroformate, NaOH, H_2O , 22 °C, 1 h, 79%; iv, SOCl₂, MeOH, 65 °C; 40 min, 74%; v, SOCl₂, pyridine, DMAP, CH_2Cl_2 , 22 °C, 2-3 days, 53%; vi, HBr, AcOH, 22 °C, 2 h; then NaOH, H_2O , 90 min, 91%

strongly acidic conditions. Finally, neutralisation of the resulting salt with NaOH gave the free base which cyclised to give the N^2 -(1-methoxycarbonyl-2-phenylethyl)triazepine **29** in 91% yield.

A preliminary attempt to prepare the N^2 , N^3 -disubstituted derivative **30**, by coupling the N^2 -substituted triazepine **29** with Z-alanine using isobutyl chloroformate resulted in the required material, in low yield. Nevertheless, it is evident that both N-2 and N-3 of this new fused diazepinedione system can be elaborated to give constrained *cis*-peptidyl proline peptide mimetics of defined stereochemistry and sequence.

Experimental

Elemental microanalyses were performed in the departmental microanalytical laboratory. NMR spectra were recorded on a Bruker AM-300 (1H, 300 MHz; 13C, 74.76 MHz), a Varian Gemini 200 (¹H, 200 MHz; ¹³C, 50.31 MHz), a Varian Gemini 300 (¹H, 300 MHz; ¹³C, 75.4 MHz) or by the SERC service at Warwick using a Bruker AM-400 (¹H, 400 MHz; ¹³C, 100 MHz) spectrometer. Chemical shifts are described in parts per million downfield from SiMe₄ and are reported consecutively as position ($\delta_{\rm H}$ or $\delta_{\rm C}$), relative integral, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, sep = septet, m = multiplet and br = broad), coupling constant (J/Hz) and assignment (numbering according to the IUPAC nomenclature for the compound). ¹H NMR spectra were referenced internally on ²HOH (δ 4.68), CHCl₃ (δ 7.27) or (C²H₃)₂-SO (δ 2.47). ¹³C NMR spectra were referenced on CH₃OH (δ 49.3), $C^{2}HCl_{3}$ (δ 77.5) or $(C^{2}H_{3})_{2}SO$ (δ 39.7).

Pyrrolidine ring carbons and hydrogens are assigned in NMR spectra as α , β , γ and δ , going anticlockwise from the ring nitrogen, according to normal convention. Where more than one conformational isomer can be seen in the NMR spectrum due to the presence of a tertiary amide bond, these are assigned as *c* (*cis*) or *t* (*trans*), according to the isomeric state of the

amide bond. Where two sets of peaks arise in NMR spectra due to different conformations of a constrained seven-membered ring, these are assigned as *A* and *B*, with *A* being the major isomer.

IR spectra were recorded on a Perkin-Elmer 1710 FT-IR spectrometer. The samples were prepared as Nujol mulls, solutions in chloroform or thin films between sodium chloride discs. The frequencies (ν) as absorption maxima are given in wavenumbers (cm⁻¹) relative to a polystyrene standard. Mass spectra and accurate mass measurements were recorded on a VG 70-250 SE, a Kratos MS-50 or by the SERC service at Swansea using a VG AZB-E. Fast atom bombardment spectra were recorded using glycerol as matrix. Major fragments are given as percentages of the base peak intensity (100%). UV spectra were recorded on Pye-Unicam SP8-500 or SP8-100 spectrophotometers.

Flash chromatography was performed according to the method of Still *et al.*²⁴ using Sorbsil C 60 (40–60 μ m mesh) silica gel. Analytical thin layer chromatography was carried out on 0.25 mm pre-coated silica gel plates (Macherey-Nagel SIL g/UV₂₅₄) and compounds were visualised using UV fluorescence, iodine vapour, ethanolic phosphomolybdic acid or ninhydrin.

Melting points were taken on an Electrothermal melting point apparatus and are uncorrected. Optical rotations were measured at 22 °C on an Optical Activity AA-1000 polarimeter using 10 or 20 cm path length cells and are given in units of $10^{-1} \text{ deg cm}^2 \text{ g}^{-1}$.

The solvents used were either distilled or of analar quality and light petroleum refers to that portion boiling between 40– 60 °C. Solvents were dried according to literature procedures. Ethanol and methanol were dried using magnesium turnings. DMF, propan-2-ol, 4-methylbutan-1-ol, toluene, CH_2Cl_2 , acetonitrile, diisopropylamine, triethylamine and pyridine were distilled over CaH_2 . THF and diethyl ether were dried over sodium-benzophenone and distilled under nitrogen. Thionyl chloride was distilled over sulfur and the initial fractions were always discarded. *N*-Methylmorpholine was distilled over ninhydrin.

(2.5)-N-Bromoacetylproline methyl ester 9 ($\mathbb{R}^1 = \mathbb{R}^2 = \mathbb{H}$)

To a solution of N-methylmorpholine (1.12 cm³, 10 mmol) in dry THF (25 cm³) was added bromoacetic acid **8** ($R^1 = R^2 = H$) (1.39 g, 10 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (1.36 cm³, 10 mmol) was added with stirring and the resulting suspension was stirred at -15 °C for 2 min. A mixture of (2.S)-proline methyl ester hydrochloride 25 7 (1.66 g, 10 mmol) and N-methylmorpholine (1.12 cm³, 10 mmol) in dry DMF (5 cm³) was then added in one portion to the cold suspension. The reaction mixture was allowed to warm to room temperature and was then stirred overnight. The hydrochloride salts were filtered off and the solvents were removed under reduced pressure. The resulting brown oil was dissolved in CH_2Cl_2 (50 cm³), washed with 0.5 mol dm⁻³ HCl (2 × 20 cm³) and 5% aqueous sodium hydrogen carbonate $(2 \times 20 \text{ cm}^3)$, dried (MgSO₄) and the solvent removed under reduced pressure to yield the product as a pale yellow oil. Purification by silica column chromatography using ethyl acetate-light petroleum (1:1) as the eluent gave the pure product as a clear oil (2.07 g, 83%) (HRMS: found [M + H]⁺, 250.0080. C₈H₁₃⁷⁹BrNO₃ requires 250.0079); $[a]_{D}^{22}$ -96.7 (c 1.0 in MeOH); v_{max} (thin film)/cm⁻¹ 1741 (ester CO), 1655 (amide CO) and 1448 (C–O); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.81–2.28 (*t* and *c*, 4 H, m, βCH₂ and γCH₂), 3.48-3.73 (t and c, 2 H, m, δCH₂), 3.57 (t, 3 H, s, OCH₃), 3.62 (c, 3 H, s, OCH₃), 3.73 (c, 2 H, d, J 1.4, COCH₂), 3.96 (t, 2 H, d, J 1.4, COCH₂), 4.33 (t, 1 H, dd, J₁ 4.0, J₂ 8.2, αCH) and 4.45 (c, 1 H, dd, J₁ 3.4, J₂ 7.2, αCH); $\delta_{\rm C}(50.31 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 22.63 (c, γCH_2), 25.24 (t, γCH_2), 29.47 (t, \(\beta\)CH2), 31.55 (c, \(\beta\)CH2), 42.38 (t and c, CH2Br), 47.44 (t and c, δCH_2), 52.65 (t, OCH₃), 53.32 (c, OCH₃),

59.61 (*t* and *c*, α CH), 165.49 (*t*, NCO), 165.59 (*c*, NCO) and 172.56 (*t* and *c*, *C*O₂CH₃); *m/z* (EI) 252 and 250 (3%, M⁺), 170 (40, [M - Br]⁺), 156 (2, [M - Br - CH₃ + H]⁺), 139 (14, [M - Br - OCH₃]⁺) and 70 (100, [C₄H₈N]⁺).

(9a.S)-2,3,4,5,7,8,9,9a-Octahydro-1*H*-pyrrolo[2,1-*d*][1,2,5]-triazepine-1,5-dione 6 ($\mathbb{R}^1 = \mathbb{R}^2 = \mathbb{H}$)

To a solution of hydrazine hydrate (1.5 g, 30 mmol) in ethanol (50 cm³) was added (2S)-N-bromoacetylproline methyl ester 9 $(R^1 = R^2 = H)$ (2.50 g, 10 mmol). The resulting solution was refluxed for 2 h and was then allowed to cool to room temperature whereupon colourless crystals of the product were formed (1.43 g, 57%), mp 265-266 °C (HRMS: found $[M + H]^+$, 170.0930. C₇H₁₂N₃O₂ requires 170.0930); $[a]_D^{22} - 74.4$ (c 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3403 (NH), 1664 (secondary amide CO) and 1637 (tertiary amide CO); $\delta_{\rm H}(200 \text{ MHz}; {}^{2}\text{H}_{2}\text{O})$ 1.83–2.25 (3 H, m, γCH_2 and $\frac{1}{2}\beta CH_2$), 2.41 (1 H, m, $\frac{1}{2}\beta CH_2$), 3.42-3.71 (2 H, m, δCH_2), 4.11 (1 H, d, J16.5, $\frac{1}{2}COCH_2$), 4.45-4.59 (1 H, m, α CH) and 4.59 (1 H, d, J16.5, $\frac{1}{2}COCH_2$); $\delta_H(200$ MHz; $[^{2}H_{6}]$ DMSO) 1.77–2.10 (A and B, 3 H, m, γ CH₂ and ¹₂βCH₂), 2.14-2.41 (A and B, 1 H, m, ¹₂βCH₂), 3.34-3.62 (A and B, 2 H, m, δCH₂), 3.89 (A, 1 H, d, J15.1, ¹/₂COCH₂), 4.11 (B, 1 H, d, J16.0, ¹/₂COCH₂), 4.40 (A, 1 H, d, J15.1, ¹/₂COCH₂), 4.42 (*A* and *B*, 1 H, m, αCH) and 4.61 (*B*, 1 H, d, *J* 16.0, ¹/₂COCH₂); δ_c(50.31 MHz; ²H₂O) 24.71 (A, γCH₂), 24.85 (B, γCH₂), 30.56 (B, βCH₂), 30.72 (A, βCH₂), 48.39 (A, COCH₂), 48.47 (B, COCH2), 54.51 (A, SCH2), 54.90 (B, SCH2), 61.44 (A, aCH), 61.59 (B, αCH), 166.61 (A, COCH₂), 166.80 (B, COCH₂), 170.47 (B, CONH) and 170.96 (A, CONH); δ_c(50.31 MHz; [²H₆]DMSO) 25.99 (A, γCH₂), 26.11 (B, γCH₂), 31.89 (B, βCH₂), 32.12 (A, βCH₂), 48.75 (A and B, COCH₂), 56.32 (B, δCH₂), 56.53 (A, δCH₂), 61.78 (A, αCH), 61.88 (B, aCH), 165.91 (A, COCH₂), 166.17 (B, COCH₂), 169.66 (B, CONH) and 170.44 (A, CONH); m/z (CI) 170 (100%, $[M + H]^+$), 155 (12, $[M - NH]^+$), 137 (7, $[M - 2NH_3]^+$) and 70 $(5, [C_4H_8N]^+).$

(2.S, 2' S)-N-(2'-Bromopropionyl)proline methyl ester 9a $(\mathbf{R}^1 = \mathbf{M}\mathbf{e}, \mathbf{R}^2 = \mathbf{H})$

To a stirred solution of (2S)-bromopropionic acid²⁶ **8a** (1.53 g, 10 mmol) in dry THF (20 cm³) at -40 °C was added N-methylmorpholine (1.12 cm³, 10 mmol). Isobutyl chloroformate (1.36 cm³, 10 mmol) was immediately added and the suspension stirred at -40 °C for 2 min. A mixture of (2*S*)-proline methyl ester hydrochloride 5 (1.66 g, 10 mmol) and N-methylmorpholine (1.12 cm³, 10 mmol) in dry DMF (5 cm³) was then added. The reaction mixture was allowed to warm to room temperature and stirred for a further 1 h. The hydrochloride salts were filtered off and the solvents removed under reduced pressure. The resulting clear oil was dissolved in CH₂Cl₂ (25 cm³) and washed with 0.5 mol dm⁻³ HCl (2×15 cm³) and 5% aqueous sodium carbonate $(2 \times 15 \text{ cm}^3)$. The organic phase was dried (MgSO₄) and the solvent removed under reduced pressure to yield a white solid. Recrystallisation from diethyl ether-light petroleum gave the product as colourless crystals (2.19 g, 83%), mp 116-118 °C (Found: C, 41.85; H, 5.4; N, 5.35. C₉H₁₄BrNO₃ requires C, 42.0; H, 5.35; N, 5.3%) (HRMS: found [M + H]⁺, 264.0235. C₉H₁₅⁷⁹BrNO₃ requires 264.0235); $[a]_{D}^{22}$ +127.3 (*c* 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 1752 (ester CO), 1648 (amide CO) and 1197 (C–O); δ_{H} (400 MHz; C²HCl₃) 1.76 (t, 3 H, d, J 6.6, CHCH₃), 1.77 (c, 3 H, d, J 6.7, CHCH₃), 1.80-2.28 (t and c, 4 H, m, β CH₂ and γ CH₂), 3.45-3.64 (t and c, 1 H, m, ¹/₂ & CH₂), 3.69 (t, 3 H, s, OCH₃), 3.73 (c, 3 H, s, OCH₃), 3.79–3.86 (t and c, 1 H, m, ¹/₂δCH₂), 4.18 (t, 1 H, q, J6.6, CHCH₃), 4.40 (c, 1 H, q, J6.6, CHCH₃), 4.45 (t, 1 H, dd, J_1 4.1, J_2 8.6, α CH) and 4.61 (*c*, 1 H, dd, J_1 4.1, J_2 8.6, α CH); $\delta_{\rm C}(100 \text{ MHz}; \text{C}^2\text{HCl}_3) 20.82 (t, \text{CH}C\text{H}_3), 20.97 (c, \text{CH}C\text{H}_3),$ 21.53 (c, YCH2), 24.56 (t, YCH2), 28.82 (t, BCH2), 30.77 (c, βCH₂), 39.25 (t, CHCH₃), 39.97 (c, CHCH₃), 46.69 (c, δCH₂), 46.81 (t, δCH₂), 52.03 (t, OCH₃), 52.65 (c, OCH₃), 58.99 (*t*, α CH), 59.05 (*c*, α CH), 167.56 (*t*, *C*OCHCH₃), 167.99 (*c*, *C*OCHCH₃), 171.87 (*t*, *C*O₂CH₃) and 172.05 (*c*, *C*O₂CH₃); *m/z* (FAB) 264 and 266 (100%, M⁺), 204 and 206 (38, [M - HCOOCH₃]⁺), 184 (35, [M - Br]⁺) and 128 (62, [M - COCHCH₃Br]⁺).

(2.5,2' R)-N-(2'-Bromopropionyl) proline methyl ester 9b (R¹ = H, R² = Me)

This compound was prepared in a manner identical with that for the (2S,2'S) methyl ester **9a**, using (2R)-bromopropionic acid 8b (1.53 g, 10 mmol) to give the product as colourless crystals (2.09 g, 79%), mp 113-115 °C (Found: C, 41.95; H, 5.35; N, 5.3. C₉H₁₄BrNO₃ requires C, 42.0; H, 5.35; N, 5.3%) (HRMS: found $[M + H]^+$, 264.0232. $C_9H_{15}^{-79}BrNO_3$ requires 264.0235); $[a]_{\rm D}^{22}$ +113.6 (*c* 1.0 in MeOH); $v_{\rm max}$ (Nujol)/cm⁻¹1752 (ester CO), 1646 (amide CO) and 1197 (C–O); $\delta_{\rm H}$ (400 MHz; C²HCl₃) 1.62 (c, 3 H, d, J 6.4, CHCH₃), 1.64 (t, 3 H, d, J 6.6, CHCH₃), 1.82-2.31 (t and c, 4 H, m, β CH₂ and γ CH₂), 3.44-3.66 (t and c, 1 H, m, ½8CH2), 3.71 (t, 3 H, s, OCH3), 3.75 (c, 3 H, s, OCH₃), 3.82–3.94 (t and c, 1 H, m, $\frac{1}{2}\delta$ CH₂), 4.22 (c, 1 H, q, J 6.5, CHCH₃), 4.44 (t, 1 H, q, J 6.6, CHCH₃), 4.47 (t, 1 H, dd, J₁ 4.3, J₂ 8.6, αCH) and 4.69 (c, 1 H, dd, J₁ 4.0, J₂ 6.7, αCH); $\delta_{\rm C}(100 \text{ MHz}; \text{ C}^2\text{HCl}_3) 20.21 (t, \text{CH}_{\rm CH}_3), 20.35 (c, \text{CH}_{\rm CH}_3),$ 22.21 (c, γCH₂), 24.67 (t, γCH₂), 28.82 (t, βCH₂), 30.89 (c, βCH₂), 39.27 (t, CHCH₃), 40.02 (c, CHCH₃), 46.70 (c, δCH₂), 46.75 (t, \deltaCH₂), 50.69 (t, OCH₃), 50.81 (c, OCH₃), 58.87 (c, αCH), 59.04 (t, αCH), 167.44 (t, COCHCH₃), 167.79 (c, COCHCH₃), 171.92 (t, CO₂CH₃) and 172.24 (c, CO₂CH₃); m/z (FAB) 264 and 266 (95%, M⁺), 204 and 206 (62, $[M - HCOOCH_3]^+$, 184 (41, $[M - Br]^+$) and 128 (100, $[M - COCHCH_3Br]^+$).

(4*R*,9a*S*)-4-Methyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 6a

To a solution of hydrazine hydrate (0.60 g, 12 mmol) in ethanol (20 cm^3) was added (2S, 2'S) - N - (2' - bromopropionyl) proline methyl ester 9a (2.64 g, 10 mmol). The resulting solution was refluxed for 16 h and was then allowed to cool slowly. The hydrazine hydrobromide precipitate was removed by filtration and the filtrate was concentrated under reduced pressure to give the required product as a colourless oil (1.81 g, 98%) (HRMS: found $[M + H]^+$, 184.1083. C₈H₁₄N₃O₂ requires 184.1086); $[a]_D^{22}$ +39.0 (c 1.0 in MeOH); v_{max} (thin film)/cm⁻¹ 1665 (secondary amide CO) and 1639 (tertiary amide CO); $\delta_{\rm H}$ (200 MHz; [²H₄]methanol) 1.53 (3 H, d, J7.0, CHCH₃), 1.86-2.15 (3 H, m, γCH_2 and $\frac{1}{2}\beta CH_2$), 2.39 (1 H, m, $\frac{1}{2}\beta CH_2$), 3.44–3.72 (2 H, m, δCH₂), 4.12 (1 H, q, J7.0, CHCH₃) and 4.38 (1 H, dd, J₁ 7.5, J_2 7.5, α CH); $\delta_{\rm C}(50.31$ MHz; $[{}^{2}{\rm H}_{4}]$ methanol) 17.01 (CHCH₃), 23.65 (γCH₂), 30.57 (βCH₂), 46.97 (δCH₂), 59.60 (αCH), 63.69 (CHCH₃), 167.85 (COCHCH₃) and 168.49 (CONH); m/z (CI) 184 (100%, $[M + H]^+$), 169 (20, $[M - CH_3 + H]^+$), 154 (8, $[M - NHCH_3 + H]^+$), 138 (4, $[M - NHNH - CH_3]^+$), 126 (5, $[M - NHNHCHCH_3 + H]^+$) and 70 (10, $[C_4H_8N]^+$).

(4*S*,9a*S*)-4-Methyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 6b

This compound was prepared in a manner identical with that for the (4*R*,9a*S*) diastereomer **6a**, using (2*S*,2'*R*)-*N*-(2'-bromopropionyl)proline methyl ester **9b** (2.64 g, 10 mmol) to give the required material as a colourless oil (1.75 g, 96%) (HRMS: found $[M + H]^+$, 184.1082. $C_8H_{14}N_3O_2$ requires 184.1086); $[a]_D^{22} + 28.6$ (*c* 1.0 in MeOH); v_{max} (thin film)/cm⁻¹ 1658 (secondary amide CO) and 1645 (tertiary amide CO); δ_H (200 MHz; $[^2H_4]$ methanol) 1.72 (3 H, d, *J* 6.8, CHC*H*₃), 1.93–2.29 (3 H, m, γ CH₂ and $\frac{1}{2}\beta$ CH₂), 2.52 (1 H, m, $\frac{1}{2}\beta$ CH), 3.54–3.86 (2 H, m, δ CH₂), 4.35 (1 H, q, *J* 6.8, CHCH₃) and 4.38 (1 H, m, α CH); δ_c (50.31 MHz; $[^2H_4]$ methanol) 18.36 (CH*C*H₃), 24.72 (γ CH₂), 31.43 (β CH₂), 48.55 (δ CH₂), 61.24 (α CH), 64.54 (*C*HCH₃), 167.49 (*C*OCHCH₃) and 170.68 (CONH); *m*/*z* (CI) 184 (100%, [M + H]⁺), 169 (31, [M - CH₃ + H]⁺), 154 (9, [M - NHNH-

(3*R*,8a*S*)-3-Methyl-1,2,3,4,6,7,8,8a-octahydropyrrolo[1,2-*a*]pyrazine-1,4-dione 12b

To a solution of (2*S*,2'*R*)-*N*-(*N*-benzyloxycarbonylalanyl)prolinamide 14b (0.64 g, 2 mmol) in methanol (30 cm³) was added 10% palladium on activated charcoal (30 mg) and the vessel flushed with hydrogen gas. The resulting suspension was stirred vigorously under an atmosphere of hydrogen for 12 h. The mixture was then filtered through a pre-washed Celite pad and the solvent removed under reduced pressure to give a white solid which was recrystallised from methanol to give the product as colourless crystals (0.32 g, 93%), mp 128-130 °C (lit.,²⁷ 127-129 °C) (Found: C, 57.4; H, 7.35; N, 16.65; M⁺, 168.1902. Calc. for $C_8H_{12}N_2O_2$: C, 57.15; H, 7.2; N, 16.65%; M^+ , 168.1899); $[a]_{D}^{22} = -175.2$ (*c* 1.0 in EtOH) [lit.,²⁵ - 182.3 (*c* 1.0 in EtOH)]; v_{max} (Nujol)/cm⁻¹ 3213 and 3162 (NH), 1687 (tertiary amide CO) and 1635 (secondary amide CO); $\delta_{\rm H}$ (200 MHz; [²H₄]methanol) 1.41 (3 H, d, J 7.0, CH₃), 1.84–2.13 (3 H, m, γCH₂ and ¹/₂βCH₂), 2.63–2.84 (1 H, m, ¹/₂βCH₂), 3.42–3.70 (2 H, m, δCH₂), 3.92 (1 H, q, J7.1, CHCH₃) and 4.28 (1 H, dd, J₁ 6.5, J_2 9.7, α CH); $\delta_{\rm C}$ (74.76 MHz; [²H₄]methanol) 20.19 (CH₃), 23.23 (γCH₂), 30.21 (βCH₂), 46.87 (δCH₂), 54.71 (CHCH₃), 59.46 (aCH), 169.34 (COCHCH₃) and 171.39 (CONH); m/z (EI) 168 (71%, M⁺), 140 (8, [M - CHCH₃]⁺), 125 (28, [M - NH-CHCH₃]⁺), 112 (8, [M - COCHCH₃]⁺), 97 (41, [M - CONH- $CHCH_{3}^{(-)}$ and 70 (100, $[C_{4}H_{8}N]^{+}$).

(3*S*,8a*S*)-3-Methyl-1,2,3,4,6,7,8,8a-octahydropyrrolo[1,2-*a*]-pyrazine-1,4-dione 12a

This compound was prepared in a manner identical with that for the octahydropyrrolo[1,2-a]pyrazine-1,4-dione 12b, using (2S,2'S)-N-(N-benzyloxycarbonylalanyl)prolinamide 14a (0.638 g, 2 mmol) to give a white solid which was recrystallised from methanol to give the product as colourless crystals (0.31 g, 91%), mp 152-156 °C (lit.,²⁷ 153-156 °C) (Found: C, 57.3; H, 7.25; N, 16.7; M⁺, 168.1900. Calc. for C₈H₁₂N₂O₂: C, 57.15; H, 7.2; N, 16.65%; M⁺, 168.1899); [a]²²_D -158.4 (c 1.0 in EtOH) [lit.,²⁵ –160.0 (*c* 1.0 in EtOH)]; v_{max} (Nujol)/cm⁻¹ 3287 (NH), 1687 (tertiary amide CO) and 1654 (secondary amide CO); δ_H(200 MHz; C²HCl₃) 1.41 (3 H, d, J6.6, CH₃), 1.68–2.39 (4 H, m, β CH₂ and γ CH₂), 3.50 (2 H, m, δ CH₂), 4.08 (2 H, m, α CH and CHCH₃) and 7.42 (1 H, s, NH); $\delta_{\rm C}$ (74.76 MHz; [²H₄]methanol) 16.05 (CH₃), 23.89 (γ CH₂), 29.43 (β CH₂), 46.48 (δCH₂), 52.33 (CHCH₃), 60.70 (αCH), 169.21 (COCHCH₃) and 172.78 (CONH); m/z (EI) 168 (43%, M⁺), 140 (18, [M - NHCH₃ + 2H]⁺), 125 (35, [M - NHCHCH₃]⁺), 97 (34, $[M - CONHCHCH_3]^+$) and 70 (100, $[C_4H_8N]^+$).

(2S)-N-(N-Benzyloxycarbonylglycyl) proline methyl ester 13

To a solution of N-benzyloxycarbonylglycine (2.09 g,10 mmol) in dry THF (20 cm³) was added *N*-methylmorpholine (1.12 cm³, 10 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (1.36 cm³, 10 mmol) was added with stirring and the solution was stirred at -15 °C for 2 min. A solution of (2*S*)proline methyl ester hydrochloride 7 (1.66 g, 10 mmol) and Nmethylmorpholine (1.12 cm³, 10 mmol) in dry DMF (5 cm³) was then added. The reaction mixture was allowed to warm to room temperature and stirred for 2 h. The hydrochloride salts were filtered off and the solvents removed under reduced pressure. The resulting oily residue was dissolved in CH₂Cl₂ (25 cm³) and washed with 0.5 mol dm⁻³ HCl (2×15 cm³) and 5% aqueous sodium carbonate $(2 \times 15 \text{ cm}^3)$. The organic phase was then dried (MgSO₄) and the solvent removed under reduced pressure to give a pale yellow oil. The crude material was purified by silica column chromatography using ethyl acetate-light petroleum (1:1) as the eluent to give the product as a colourless oil (2.42 g, 76%) (HRMS: found $[M + H]^+$, 321.1453. $C_{16}H_{21}N_2O_5$ requires 321.1450); $[a]_{D}^{22}$ -68.1 (*c* 1.0 in MeOH); v_{max} (thin film)/ cm⁻¹ 3336 (NH), 1732 (ester CO), 1711 (carbamate CO), 1652 (amide CO) and 736 and 699 (aromatic CH); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.81-2.33 (t and c, 4 H, m, βCH₂ and γCH₂), 3.35-3.69 (t and c, 2 H, m, δCH_2), 3.73 (t, 3 H, s, OCH_3), 3.76 (c, 3 H, s, OCH₃), 4.02 (t and c, 2 H, d, J2.6, COCH₂), 4.40 (t, 1 H, dd, J₁ 4.0, J₂ 6.6, αCH), 4.53 (c, 1 H, dd, J₁ 3.8, J₂ 7.0, αCH), 5.12 (t and c, 2 H, s, CH₂Ph), 5.74 (t and c, 1 H, t, J 2.6, NHCH₂) and 7.35 (t and c, 5 H, s, Ar–H); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 22.52 (c, γCH₂), 24.97 (t, γCH₂), 29.33 (t, βCH₂), 31.66 (c, βCH₂), 43.52 (c, COCH₂), 43.66 (t, COCH₂), 46.20 (t, \deltaCH₂), 47.00 (c, δCH₂), 52.68 (t, OCH₃), 53.15 (c, OCH₃), 58.80 (c, αCH), 59.20 (t, αCH), 128.28 (t and c, Ar-CH para), 128.37 (t and c, Ar-CH ortho), 128.81 (t and c, Ar-CH meta), 136.87 (t and c, Ar-C quaternary), 156.63 (t and c, CO₂CH₂Ph), 167.41 (t, CO₂CH₂), 167.69 (c, CO₂CH₂), 172.29 (c, CO₂CH₃) and 172.68 (t, CO_2CH_3 ; m/z (CI) 321 (100%, $[M + H]^+$), 230 (10, $[M - H]^+$) $CH_2Ph + H]^+$), 213 (28, $[M - OCH_2Ph]^+$), 187 (36, $[M - OCH_2Ph]^+$) $CO_2CH_2Ph + 2H]^+$), 157 (63, $[M - CO_2CH_2Ph - OCH_2 +$ 3H]⁺) and 91 (41, [CH₂Ph]⁺).

(2*S*,2′*S*)-*N*-(*N*-Benzyloxycarbonylalanyl)proline methyl ester 13a

This compound was prepared in a manner identical with that for the (2S)-N-(N-benzyloxycarbonylglycyl)proline methyl ester 13, using (2.S)-N-benzyloxycarbonylalanine (2.23 g, 10 mmol) instead of N-benzyloxycarbonylglycine to give a yellow oil which was purified by silica chromatography using ethyl acetate-light petroleum (1:1) as the eluent to give the product as a clear oil (2.89 g, 84%) (HRMS: found [M + H]⁺, 335.1605. $C_{17}H_{23}N_2O_5$ requires 335.1606); $[a]_D^{22}$ -86.5 (c 1.0 in MeOH); v_{max}(Nujol)/cm⁻¹ 2982 and 2953 (NH), 1746 (ester CO), 1716 (carbamate CO), 1654 (amide CO) and 669 and 666 (aromatic CH); $\delta_{\rm H}(200 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 1.11 (*t* and *c*, 3 H, d, *J* 6.6, CHCH₃), 1.49–2.02 (t and c, 4 H, m, β CH₂ and γ CH₂), 3.21– 3.45 (t and c, 2 H, m, δCH_2), 3.40 (t and c, 3 H, s, OCH_3), 4.24 (t and c, 1 H, m, aCH), 4.26 (t and c, 1 H, q, J6.6, CHCH₃), 4.82 (t and c, 2 H, s, CH, Ph), 5.97 (c, 1 H, d, J8.6, NH), 6.13 (t, 1 H, d, J 7.8, NH) and 7.06 (t and c, 5 H, s, Ar-H); $\delta_{\rm C}$ (50.31 MHz; $C^{2}HCl_{3}$) 18.03 (t, CHCH₃), 19.06 (c, CHCH₃), 22.91 (c, γ CH₂), 25.14 (t, YCH2), 29.09 (t, BCH2), 31.47 (c, BCH2), 46.75 (c, δCH₂), 46.96 (t, δCH₂), 48.56 (t, CHCH₃), 48.77 (c, CHCH₃), 52.30 (t, OCH₃), 53.87 (c, OCH₃), 58.97 (t, aCH), 59.29 (c, αCH), 66.67 (t, CH₂Ph), 67.05 (c, CH₂Ph), 128.14 (t and c, Ar-CH ortho), 128.68 (t and c, Ar-CH para), 129.36 (t and c, Ar-CH meta), 137.05 (t and c, Ar-C quaternary), 155.73 (c, CO₂CH₂Ph), 156.13 (t, CO₂CH₂Ph), 171.65 (t and c, COCHCH₃), 172.51 (c, CO₂CH₃) and 172.62 (t, CO₂CH₃); m/z (CI) 335 (100%, $[M + H]^+$), 319 (7, $[M - CH_3]^+$), 291 $(13, [M - COCH_3]^+), 227 (12, [M - OCH_2Ph]^+) and 91 (8,$ [PhCH,]⁺).

Catalytic removal of the Z protecting group of this compound gave the diazepinedione **12a**, identical to the material prepared from compound **6a**.

(2.S, 2'R)-N-(N-Benzyloxycarbonylalanyl)proline methyl ester 13b

This compound was prepared in a manner identical with that for the methyl ester **13**, using (2*R*)-*N*-benzyloxycarbonylalanine (2.23 g, 10 mmol) instead of (2*S*)-*N*-benzyloxycarbonylalanine to give a yellow oil which was purified by silica column chromatography using ethyl acetate–light petroleum (1:1) as the eluent to give the product as a colourless oil (2.40 g, 72%) (HRMS: found [M + H]⁺, 335.1614. C₁₇H₂₃N₂O₅ requires 335.1607); [*a*]_D²² –22.9 (*c* 1.0 in MeOH); ν_{max} (thin film)/cm⁻¹ 2982 and 2955 (NH), 1745 (ester CO), 1719 (carbamate CO), 1652 (amide CO) and 736 and 699 (aromatic CH); δ_{H} (200 MHz; C²HCl₃) 1.27 (*c*, 3 H, d, *J*7.0, CHCH₃), 1.33 (*t*, 3 H, d, *J*6.6, CHCH₃), 1.79–2.33 (*t* and *c*, 4 H, m, β CH₂ and γ CH₂), 3.38–3.62 (*t* and *c*, 2 H, m, δ CH₂), 3.70 (*t*, 3 H, s, OCH₃), 3.74 (*c*, 3 H, s, OCH₃), 4.31 (*c*, 1 H, dq, *J*₁ 7.0, *J*₂ 8.0, CHCH₃), 4.42 (*t*, 1 H, dd, *J*₁ 2.1, *J*₂ 9.1,

αCH), 4.54 (t, 1 H, dq, J₁ 6.6, J₂ 7.6, CHCH₃), 4.91 (c, 1 H, dd, J₁ 3.1, J₂ 7.3, αCH), 5.08 (t and c, 2 H, s, CH₂Ph), 5.59 (c, 1 H, d, J 8.0, NH), 5.78 (t, 1 H, d, J 7.6, NH), 7.31 (c, 5 H, s, Ar-H) and 7.32 (t, 5 H, s, Ar–H); $\delta_{\rm C}(50.31 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 17.89 (c, CHCH₃), 18.48 (t, CHCH₃), 22.67 (c, γCH₂), 24.92 (t, γCH₂), 29.22 (t, \beta CH2), 31.31 (c, \beta CH2), 46.81 (c, \delta CH2), 47.00 (t, δCH₂), 48.42 (c, CHCH₃), 48.67 (t, CHCH₃), 52.25 (t, OCH₃), 52.72 (c, OCH₃), 59.32 (t, αCH), 60.46 (c, αCH), 66.63 (t and c, CH₂Ph), 128.13 (t and c, Ar-CH), 128.64 (t and c, Ar-CH), 136.93 (c, Ar-C quaternary), 137.03 (t, Ar-C quaternary), 155.77 (t, CO₂CH₂Ph), 156.37 (c, CO₂CH₂Ph), 171.05 (t, COCHCH₃), 171.29 (c, COCHCH₃), 172.53 (t, CO₂CH₃) and 172.98 (t, CO_2CH_3); m/z (CI) 335 (100%, $[M + H]^+$), 230 (5, $[M - CH_2Ph - CH_3 + 2H]^+)$, 201 (3, $[M - CO_2CH_2Ph +$ 2H]⁺), 196 (4, $[M - OCH_2Ph - OCH_3]^+$) and 169 (6, $[M - OCH_3]^+$) $CO_2CH_2Ph - OCH_3 + H]^+$).

(2S)-N-(N-Benzyloxycarbonylglycyl)prolinamide 14

To a saturated solution of ammonia in dry methanol (20 cm³) was added (2*S*)-*N*-(*N*-benzyloxycarbonylglycyl)proline methyl ester 13 (1.60 g, 5 mmol). The solution was placed in a tightly stoppered vessel and left at room temperature. When the reaction was complete as judged by TLC (9 days), the stoppered vessel was cooled to 0 °C, opened and nitrogen gas bubbled through at room temperature for 5 min. The solvent was then removed under reduced pressure to give a white solid which was recrystallised from methanol-diethyl ether to give the product as a white solid (1.24 g, 81%), mp 148-149 °C (Found: C, 59.15; H, 6.4; N, 13.8. C₁₅H₁₉N₃O₄ requires C, 59.0; H, 6.3; N, 13.75%) (HRMS: found $[M + H]^+$, 306.1459. $C_{15}H_{20}N_3O_4$ requires 306.1454); $[a]_{D}^{22}$ -60.1 (*c* 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3394 and 3213 (NH), 1687 (carbamate CO), 1671 (tertiary amide CO),1639 (primary amide CO) and 753 and 701 (aromatic CH); $\delta_{\rm H}(200 \text{ MHz}; [{}^{2}\text{H}_{6}]\text{DMSO}) 1.67-2.34 (t \text{ and } c, 4 \text{ H}, \text{ m}, \gamma \text{CH}_{2}$ and (3CH2), 3.31-3.69 (t and c, 2 H, m, (3CH2), 3.79 (c, 2 H, d, J 5.8, COCH₂), 3.86 (t, 2 H, d, J5.8, COCH₂), 4.21 (t, 1 H, dd, J₁ 5.8, J, 7.6, aCH), 4.35 (c, 1 H, dd, J₁ 2.6, J₂ 8.2, aCH), 5.01 (t and c, 2 H, s, CH₂Ph), 7.01 (t, 1 H, s, ¹/₂CONH₂), 7.28 (c, 1 H, s, ¹/₂CONH₂), 7.32 (t, 1 H, s, ¹/₂CONH₂), 7.38 (t and c, 5 H, s, Ar-H) and 7.61 (c, 1 H, s, $\frac{1}{2}$ CONH₂); δ_{C} (50.31 MHz; [²H₄]methanol) 23.62 (c, YCH₂), 25.90 (t, YCH₂), 31.05 (t, BCH₂), 33.69 (c, βCH₂), 44.35 (t, COCH₂), 44.46 (c, COCH₂), 47.83 (t, δCH₂), 48.61 (c, δCH₂), 61.26 (c, αCH), 61.86 (t, αCH), 68.01 (t and c, CH₂Ph), 129.16 (t and c, Ar-CH para), 129.28 (t and c, Ar-CH ortho), 129.74 (t and c, Ar-CH meta), 138.43 (t and c, Ar-C quaternary), 159.30 (t and c, CO₂CH₂Ph), 170.58 (t, COCH₂), 170.72 (c, COCH₂), 177.49 (c, CONH₂) and 174.94 (t, CONH₂); m/z (CI) 306 (33%, [M + H]⁺), 198 (100, [M - OCH₂Ph]⁺), 172 $(11, [M - CO_2CH_2Ph + 2H]^+), 155 (38, [M - CO_2CH_2Ph - CO_2CH_2Ph))$ $NH_{2} + H^{+}$, 139 (15, $[M - CO_{2}CH_{2}Ph - NH_{2} - NH^{+})$ and 91 (22, [CH₂Ph]⁺).

Upon catalytic hydrogenolysis of the N-Z protection, this compound cyclised to give **12**, see below.

(8a.5)-1,2,3,4,6,7,8,8a-Octahydropyrrolo[1,2-*a*]pyrazine-1,4dione 12

Method 1. To liquid ammonia (40 cm³) at -60 °C under an atmosphere of dry nitrogen was added a solution of the triazepinedione **6** (1.69 g, 10 mmol) in dry THF (10 cm³). The resulting solution was stirred vigorously and sodium (approx. 0.8 g, 30 mmol) was added in small pieces until a homogenous dark blue colour was obtained. This blue colour was maintained for 15 min, after which time the solution cleared and ammonium chloride (1.60 g, 30 mmol) was added. The reaction mixture was allowed to warm to room temperature, methanol (20 cm³) was added, and the mixture was left stirring for 1 h. The solvents were removed under reduced pressure and the resulting white solid was dissolved in water (25 cm³) and extracted with ethyl acetate (3 × 25 cm³). The combined organic fractions were dried (MgSO₄) and the solvent removed under

reduced pressure to give a white solid which was recrystallised from methanol to give the required material as colourless crystals (1.23 g, 80%). All analytical and spectroscopic data were identical to those described for Method 2, see below.

Method 2. To a solution of (2S)-N-(N-benzyloxycarbonylglycyl)prolinamide 14 (0.610 g, 2 mmol) in methanol (50 cm³) was added 10% palladium on activated charcoal (30 mg) and the vessel flushed with hydrogen. The resulting suspension was stirred vigorously under an atmosphere of hydrogen for 12 h. The mixture was filtered through a Celite pad and the solvent removed under reduced pressure to give a white solid which was recrystallised from methanol to give the required material as colourless crystals (0.29 g, 93%), mp 205-208 °C (lit., 19 208-210 °C) (Found: C, 54.45; H, 6.8; N, 18.4; M⁺, 154.0747. Calc. for $C_7H_{10}N_2O_2$: C, 54.5; H, 6.65; N, 18.3%; M⁺, 154.0742); $[a]_D^{22}$ -184.3 (c 0.5 in MeOH) [lit.,¹⁹ -196.5 (c 0.5 in MeOH)]; v_{max} (Nujol)/cm⁻¹ 3162 and 3110 (NH), 1681 (tertiary amide CO) and 1652 (secondary amide CO); $\delta_{\rm H}$ (300 MHz; [²H₄]methanol) 2.03–2.38 (3 H, m, γCH_2 and $\frac{1}{2}\beta CH_2$), 2.45–2.64 (1 H, m, ½βCH₂), 3.73 (2 H, m, δCH₂), 3.93 (1 H, d, J 16.8, ¹/₂COCH₂), 4.30 (1 H, d, J 16.8, ¹/₂COCH₂) and 4.42 (1 H, dd, J₁ 6.8, J_2 6.8, αCH); $\delta_{\rm C}$ (74.76 MHz; [²H₄]methanol) 23.60 (γCH₂), 29.68 (βCH₂), 46.60 (COCH₂), 47.29 (δCH₂), 60.15 (αCH), 166.74 (COCH₂) and 172.27 (CONH); m/z (EI) 154 (70%, M⁺), 126 (8, $[M - NHCH_2 + H]^+$), 111 (81, $[M - CONH]^+$), 98 (28, $[M - CONHCH_2 + 2H]^+)$, 83 (100, $[M - CO_2 - NHCH_2 +$ $2H^{+}$ and 70 (78, $[C_4H_8N^{+}]^{+}$).

(2*S*,2′*S*)-*N*-(*N*-Benzyloxycarbonylalanyl)prolinamide 14a

This compound was prepared in a manner identical with that for (2S)-N-(N-benzyloxycarbonylglycyl)prolinamide 14, starting from the (2S,2'S)-N-(N-tert-butoxycarbonylalanyl)proline methyl ester 13a (0.67 g, 2 mmol) to give an off-white solid which was recrystallised from methanol-diethyl ether to give the product as colourless crystals (0.51 g, 79%) mp 167-169 °C (Found: C, 60.5; H, 6.75; N, 13.3. $C_{16}H_{21}N_3O_4$ requires C, 60.2; H, 6.65; N, 13.2%) (HRMS: found [M + H]⁺, 320.1612. $C_{16}H_{22}N_{3}O_{4}$ requires 320.1610); $[a]_{D}^{22}$ -81.3 (c 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3402 and 3323 (NH), 1668 (carbamate CO), 1653 (tertiary amide CO), 1646 (primary amide CO) and 754 and 696 (aromatic CH); $\delta_{\rm H}$ (200 MHz; [²H₆]DMSO) 1.20 (3 H, d, J7.6, CH₃), 1.62-2.18 (4 H, m, βCH₂ and γCH₂), 3.58 (2 H, m, δCH₂), 4.22 (1 H, dd, J₁ 7.5, J₂ 2.7, αCH), 4.32 (1 H, m, CHCH₃), 5.01 (2 H, s, CH₂Ph), 6.88 (1 H, s, ¹/₂CONH₂), 7.22 (1 H, s, ¹/₂CONH₂), 7.35 (5 H, s, Ar-H) and 7.51 (1 H, d, J 7.6, NH); $\delta_{\rm C}$ (74.76 MHz; [²H₄]methanol) 17.63 (*t*, CH₃), 18.28 (*c*, CH₃), 23.52 (c, γCH₂), 26.42 (t, γCH₂), 31.17 (t, βCH₂), 33.36 (c, βCH₂), 48.93 (t and c, δCH₂), 50.36 (t, CHCH₃), 50.72 (c, CHCH₃), 61.86 (t, aCH), 62.05 (c, aCH), 68.11 (t, CH₂Ph), 68.73 (c, CH, Ph), 129.32 (t and c, Ar-CH ortho), 129.49 (t and c, Ar-CH para), 129.94 (t and c, Ar-CH meta), 138.66 (t and c, Ar-C quaternary), 158.66 (t and c, CO₂CH₂Ph), 174.45 (t and c, COCHCH₂), 176.93 (c, CONH₂) and 177.54 (t, CONH₂); m/z (CI) 320 (100%, $[M + H]^+$), 303 (20, $[M - NH_2]^+$), 291 $(21, [M - CO]^+), 276 (19, [M - CONH_2 + H]^+), 227 (34, [M - CONH_2 + H]^+))$ $PhCH_{2} + H]^{+}$, 212 (38, $[M - PhCH_{2}O]^{+}$), 91 (40, $[PhCH_{2}]^{+}$) and 70 (23, [C₄H₈N]⁺).

Under the conditions described for the preparation of compound **12**, Method 2, this compound **14a** spontaneously lactamised to give dioxopiperazine **12a**.

(2S,2' R)-N-(N-Benzyloxycarbonylalanyl)prolinamide 14b

This compound was prepared in a manner identical with that for the (2.S, 2'S)-*N*-(*N*-benzyloxycarbonylalanyl)prolinamide **14a**, using (2.S, 2'R)-*N*-(*N*-benzyloxycarbonylalanyl)proline methyl ester **13b** (1.60 g, 5 mmol) instead of (2.S, 2'S)-*N*-(*N*benzyloxycarbonylalanyl)proline methyl ester **13a** to give the product as a clear oil (1.57 g, 98%) (HRMS: found $[M + H]^+$, 320.1606. C₁₆H₂₂N₃O₄ requires 320.1610); $[a]_{D}^{22}$ –19.8 (*c* 1.0 in MeOH); v_{max} (thin film)/cm⁻¹ 1701 (carbamate CO), 1683 (ter-

tiary amide CO), 1642 (primary amide CO) and 743 and 699 (aromatic CH); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.30 (3 H, d, J 6.8, CHCH₃), 1.70-2.43 (4 H, m, βCH₂ and γCH₂), 3.22-3.60 (1 H, m, $\frac{1}{2}\delta$ CH₂), 3.62–3.97 (1 H, m, $\frac{1}{2}\delta$ CH₂), 4.23–4.61 (2 H, m, α CH and CHCH₃), 5.04 (2 H, s, CH₂Ph), 5.79 (1 H, s, ¹/₂NH₂), 6.10 (1 H, d, J 6.6, NHCHCH₃), 6.86 (1 H, s, ¹/₂NH₂) and 7.32 (5 H, s, Ar–H); δ_c(50.31 MHz; [²H₄]methanol) 17.44 (t, CH₃), 18.12 (c, CH₃), 23.96 (c, γCH₂), 25.89 (t, γCH₂), 31.08 (t, βCH₂), 33.77 (c, βCH₂), 48.42 (c, δCH₂), 48.75 (t, δCH₂), 50.04 (t and c, CHCH₃), 62.03 (c, aCH), 62.29 (t, aCH), 67.87 (c, CH₂Ph), 68.13 (t, CH₂Ph), 129.24 (t and c, Ar-CH para), 129.37 (t and c, Ar-CH ortho), 129.95 (t and c, Ar-CH meta), 138.46 (t and c, Ar-C quaternary), 158.56 (c, CO₂CH₂Ph), 158.68 (t, CO₂CH₂-Ph), 174.36 (t, COCHCH₃), 175.02 (c, COCHCH₃), 177.24 (c, CO₂NH₂) and 177.39 (*t*, CO₂NH₂); *m*/*z* (CI) 320 (100%, $[M + H]^+$), 303 (32, $[M - NH_2]^+$), 291 (15, $[M - CO]^+$), 276 $(22, [M - CONH_2 + H]^+), 230 (31, [M - Ph - CH_3 + 3H]^+),$ 186 (36, $[M - CO_2CH_2Ph + 2H]^+$), 91 (52, $[PhCH_2]^+$) and 70 $(18, [C_4H_8N]^+).$

Under the conditions described for the preparation of compound **12**, Method 2, this compound **14b** spontaneously lactamised to give dioxopiperazine **12b**.

(9a.*S*)-3-Methyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo[2,1-*d*]-[1,2,5]triazepine-1,5-dione 15

To a stirred solution of methylhydrazine (0.46 g, 10 mmol) in CH_2Cl_2 (15 cm³) was added (2*S*)-*N*-bromoacetylproline methyl ester (1.25 g, 5 mmol). The mixture was refluxed for 30 min, allowed to cool to room temperature and extracted with water $(2 \times 20 \text{ cm}^3)$. The organic phase was separated, the solvent removed under reduced pressure and the residue redissolved in ethanol (15 cm³). Methylhydrazine (0.23 g, 5 mmol) was added and the resulting solution refluxed for 1 h. Removal of the solvent under reduced pressure gave a white solid which was recrystallised from methanol to give the product as colourless crystals (0.75 g, 82%), mp 236-238 °C (Found: C, 52.3; H, 7.3; N, 22.85. C₈H₁₃N₃O₂ requires C, 52.45; H, 7.2; N, 22.95%); $[a]_{D}^{22}$ +118.1 (c 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3101 (NH), 1699 (tertiary amide CO) and 1597 (secondary amide CO); $\delta_{\rm H}(200 \text{ MHz}; [^{2}H_{4}]$ methanol), 1.76–2.15 (3 H, m, $\frac{1}{2}\beta CH_{2}$ and γCH₂), 2.33-2.58 (1 H, m, ½βCH₂), 2.69 (3 H, s, NCH₃), 3.41 (1 H, d, J 17.2, $\frac{1}{2}$ COCH₂), 3.55 (2 H, dd, $J_1 = J_2$ 6.6, δ CH₂), 3.71 (1 H, d, $J_{17.2}$, $\frac{1}{2}COCH_{2}$) and 5.19 (1 H, dd, $J_{1} = J_{2}$ 7.1, α CH); $\delta_{C}(50.31$ MHz; [²H₄]methanol) 22.63 (γ CH₂), 28.05 (βCH₂), 44.12 (NCH₃), 48.46 (δCH₂), 58.50 (COCH₂), 63.55 (aCH), 169.77 (COCH₂) and 173.36 (CONH); m/z (FAB) 184 $(100\%, [M + H]^{+}), 168 (7, [M - CH_{3}]^{+}), 154 (19, [M - CH_{3}]^{+}))$ NCH_{3}^{+} , 112 (26, $[M - CONHNCH_{3} + H]^{+}$) and 70 (63, $[C_4H_8N]^+$).

(4*R*,9a*S*)-3,4-Dimethyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 15a

This compound was prepared in a manner identical with that for the triazepine-1,5-dione 15, using (2S,2'S)-N-(2'-chloropropionyl)proline methyl ester 25a (1.10 g, 5 mmol) instead of (2S)-N-bromoacetylproline methyl ester to give the product as colourless crystals (0.83 g, 84%), mp 155-158 °C (Found: C, 54.6; H, 7.85; N, 21.4. $C_9H_{15}N_3O_2$ requires C, 54.8; H, 7.7; N, 21.3%) (HRMS: found $[M + H]^+$, 198.1244. C₉H₁₆N₃O₂ requires 198.1243); $[a]_D^{22}$ +104.2 (*c* 0.1 in MeOH); v_{max} (Nujol)/ cm⁻¹ 3184 and 3124 (NH), 1695 (tertiary amide CO) and 1600 (secondary amide CO); $\delta_{\rm H}$ (300 MHz; C²HCl₃) 1.38 (3 H, d, J 6.6, CHCH₃), 1.67-2.02 (3 H, m, γCH₂ and ½βCH₂), 2.39-2.75 (1 H, m, ¹/₂βCH₂), 2.58 (3 H, s, NCH₃), 3.37 (1 H, q, J 6.6, CHCH₃), 3.35-3.66 (1 H, m, ¹/₂δCH₂), 3.80 (1 H, m, ¹/₂δCH₂), 5.10 (1 H, dd, J₁ 5.5, J₂ 7.6, αCH) and 7.69 (1 H, s, NH); $\delta_{\rm C}(74.76 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 18.69 (CHCH₃), 22.91 (γ CH₂), 27.01 (βCH_2) , 41.80 (NCH₃), 48.76 (δCH_2), 57.04 (αCH), 68.64 (CHCH₃), 171.69 (COCHCH₃) and 173.09 (CONH); m/z (FAB) 198 (100%, $[M + H]^+$), 182 (6, $[M - CH_3]^+$), 168 (24, $[M-NCH_3]^+),\ 126\ (19,\ [M-CONHCH_3+H]^+)$ and 70 (85, $[C_4H_8N]^+).$

(4*S*,9a*S*)-3,4-Dimethyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 15b

This compound was prepared in a manner identical with that for the (4R,9aS) diastereomer 15a, using (2S,2'R)-N-(2'-chloropropanoyl)proline methyl ester 25b (1.10 g, 5 mmol) to give the product as colourless crystals (0.79 g, 80%), mp 146-150 °C (Found: C, 54.6; H, 7.9; N, 21.4. C₉H₁₅N₃O₂ requires C, 54.8; H, 7.7; N, 21.3%) (HRMS: found [M + H]⁺, 198.1244. $C_9H_{16}N_3O_2$ requires 198.1243); $[a]_D^{22}$ +18.2 (c 0.1 in MeOH); v_{max} (Nujol)/cm⁻¹ 2926 and 2854 (NH), 1696 (tertiary amide CO) and 1601 (secondary amide CO); $\delta_{\rm H}(200 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 1.31 (3 H, d, J 7.0, CHCH₃), 1.71-2.12 (3 H, m, ¹/₂βCH₂ and γCH₂), 2.38-2.62 (1 H, m, ½βCH₂), 2.67 (3 H, s, NCH₃), 3.60 (2 H, m, δ CH₂), 3.65 (1 H, q, J7.0, CHCH₃), 5.00 (1 H, dd, $J_1 = J_2$ 7.6, α CH) and 7.24 (1 H, s, NH); $\delta_{\rm C}(50.31$ MHz; C²HCl₃) 19.21 (CHCH₃), 22.70 (γCH₂), 27.69 (βCH₂), 41.88 (NCH₃), 48.57 (δCH₂), 58.10 (αCH), 65.08 (CHCH₃), 170.80 (COCHCH₃) and 172.65 (CONH); m/z (FAB) 198 (100%, [M + H]⁺), 182 $(6, [M - CH_3]^+), 168 (24, [M - NCH_3]^+), 126 (19, [M - NCH_3]^+))$ $CONHCH_3 + H]^+$) and 70 (85, $[C_4H_8N]^+$).

(4*R*,9a*S*)-3-Acetyl-4-methyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo[2,1-*d*][1,2,5]triazepine-1,5-dione 16

To a stirred solution of the triazepinedione 6a (240 mg, 1.31 mmol) in acetic acid (5 cm³, 88 mmol) was added acetic acid (140 mm³, 1.4 mmol) and pyridine (130 mm³, 1.6 mmol). After 45 h, 6 mol dm⁻³ aqueous sodium hydroxide (15 cm³, 90 mmol) was added and the slurry was extracted with CH_2Cl_2 (2 × 15 cm³) and diethyl ether $(2 \times 15 \text{ cm}^3)$. The combined organic fractions were dried (MgSO₄) and the solvent removed under reduced pressure. The crude product was purified by silica chromatography using light petroleum and then CH2Cl2ethanol (8:2) as the eluent to give the product as a colourless oil (140 mg, 48%) (HRMS: found M^+ , 225.1104. $C_{10}H_{15}N_3O_3$ requires 225.1113); $[a]_{D}^{22}$ -47.2 (c1.1 in MeOH); v_{max} (CCl₄)/cm⁻¹ 3437 (NH), 1680 (CO) and 1641 (CO); $\delta_{\rm H}(200~{\rm MHz};~{\rm C}^2{\rm HCl}_3)$ 1.47 (3 H, d, J7.2, CHCH₃), 1.85-2.09 (6 H, m, COCH₃, ¹₃βCH₂ and γCH_2), 2.39 (1 H, m, $\frac{1}{2}\beta CH_2$), 3.51-3.61 (2 H, m, δCH_2), 4.08-4.22 (2 H, m, CHCH₃ and αCH) and 8.81 (1 H, s, NH); δ_C(50.31 MHz; C²HCl₃) 16.60 (CHCH₃), 20.71 (COCH₃), 22.21 (γCH₂), 28.92 (βCH₂), 45.39 (δCH₂), 57.71 (αCH), 62.19 (CHCH₃), 165.74 (COCHCH₃), 165.82 (CONH) and 169.23 $(COCH_3); m/z$ (EI) 225 (52%, M⁺), 183 (80 [M + H -COCH₃]⁺), 167 (49, [M - COCH₃ - CH₃]⁺), 126 (78, [M + $H - CH_3CONNHCO]^+$ and 70 (100, $[C_4H_8N]^+$).

N-tert-Butoxycarbonyl-*N*-benzyloxycarbonyl-*N*-methylhydrazine

To a solution of N-benzyloxycarbonyl-N-methylhydrazine (3.60 g, 20 mmol) in propan-2-ol (25 cm³) was added a solution of di-tert-butyl dicarbonate (4.80 g, 22 mmol) in CH₂Cl₂ (10 cm³). The mixture was stirred for 4 h, and the solvent removed under reduced pressure to yield the product as a pale yellow oil. Purification by silica column chromatography using ethyl acetate-light petroleum (1:4) as the eluent gave the pure product as a clear oil (4.94 g, 88%) (HRMS: found $[M + H]^+$, 281.1504. C₁₄H₂₁N₂O₄ requires 281.1501); v_{max} (thin film)/cm⁻¹ 3307 (NH), 1733 (Z CO) and 1707 (Boc CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.29 [9 H, s, C(CH₃)₃], 2.45 (3 H, s, NCH₃), 5.12 (2 H, s, CH₂Ph), 6.12 (1 H, s, NH) and 7.31 (5 H, s, Ar–H); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 28.31 [C(CH₃)₃], 39.25 (NHCH₃), 63.98 (CH₂Ph), 80.74 [C(CH₃)₃], 128.15 (Ar-CH ortho), 128.33 (Ar-CH meta), 128.78 (Ar-CH para), 136.43 (Ar-C quaternary), 155.52 (CO₂CH₂Ph) and 157.11 (CO₂Bu⁴); m/z (CI) 281 $(54\%, [M + H]^+), 265 (32, [M - CH_3]^+), 225 (71, [M - CH_3]^+))$ $C(CH_3)_3 + 2H]^+$, 191 {57, [M - CO₂C(CH₃)₃ + 2H]⁺}, 91 (100, $[PhCH_2]^+$) and 57 {88, $[C(CH_3)_3]^+$ }.

(2S)-N-Chloroacetylproline 19

To a stirred suspension of (2S)-proline (7.50 g, 65.1 mmol) in ethyl acetate (150 cm³) was added chloroacetyl chloride (8.0 cm³, 99.7 mmol). The suspension was refluxed for 50 min and cooled to yield the product as colourless crystals (7.93 g, 64%); mp 106-108 °C (lit.,²⁸ 112 °C) (Found: C, 43.8; H, 5.2; N, 7.2. C₇H₁₀ClNO₃ requires C, 43.9; H, 5.3; N, 7.3%); [a]²²_D - 110.6 (c0.9 in H₂O) [lit.,²⁸ –114 (c 2 in H₂O)]; v_{max} (Nujol)/cm⁻¹ 2900 (OH), 1722 (acid CO), 1620 (amide CO) and 690 (C-Cl); $\delta_{\rm H}(200 \text{ MHz}; [^{2}H_{4}]\text{methanol}) 1.90-2.13 (3 \text{ H}, \text{ m}, \frac{1}{2}\beta \text{CH}_{2} \text{ and}$ γCH_2), 2.18–2.35 (1 H, m, $\frac{1}{2}\beta CH_2$), 3.51–3.74 (2 H, m, δCH₂), 4.28 (2 H, d, J 2.8, CH₂Cl), 4.41-4.48 (t, 1 H, m, α CH) and 4.62–4.69 (*c*, 1 H, m, α CH); δ_{c} (74.76 MHz; $[^{2}H_{4}]$ methanol) 23.49 (c, γCH_{2}), 25.99 (t, γCH_{2}), 30.51 (t, βCH₂), 32.37 (c, βCH₂), 43.12 (c, CH₂Cl), 43.23 (t, CH₂Cl), 48.46 (t, δCH₂), 48.73 (c, δCH₂), 60.98 (t and c, αCH), 167.76 (t, CON), 168.17 (c, CON), 174.87 (c, CO₂H) and 175.34 $(t, CO_2H); m/z$ (EI) 191 (6%, M⁺), 146 (52, $[M - CO_2H]^+),$ 112 (36, $[M - Cl - CO_2H + H]^+$), 83 (23, $C_5H_7N^+$), 70 (100, $C_4H_8N^+$) and 41 (37, COCH⁺).

N-tert-Butoxycarbonyl-*N* -methylhydrazine 18

To a solution of *N-tert*-butoxycarbonyl-*N*'-benzyloxycarbonyl-N-methylhydrazine (1.40 g, 5 mmol) in methanol (30 cm³) was added 10% palladium on activated charcoal (75 mg) and the vessel flushed with hydrogen gas. The resulting suspension was stirred vigorously under an atmosphere of hydrogen for 12 h. The mixture was then filtered through a pre-washed Celite pad, and the solvent removed under reduced pressure to give a clear oil which solidified on standing to give colourless crystals of the product (0.68 g, 93%), mp 46-49 °C (Found: C, 49.25; H, 9.9; N, 19.2. C₆H₁₄N₂O₂ requires C, 49.3; H, 9.65; N, 19.15%); v_{max}(Nujol)/cm⁻¹ 3320 (amine NH), 3241 (amide NH) and 1700 (CO); $\delta_{\rm H}(200 \,{\rm MHz};{\rm C}^2{\rm HCl}_3)$ 1.33 [9 H, s, C(CH₃)₃], 2.48 (3 H, s, NHCH₃), 3.76 (1 H, s, NHCH₃) and 6.00 (1 H, s, CONH); δ_C(50.31 MHz; C²HCl₃) 28.88 [C(*C*H₃)₃], 39.69 (NHCH₃), 80.82 [C(CH₃)₃] and 157.27 (CO₂Bu⁴); m/z (EI) 146 (15%, M⁺), 131 (3, $[M - CH_3]^+$), 103 (51, $[M - NHNHCH_3 + 2H]^+$), 90 {33, $[M - C(CH_3)_3 + H]^+$, 73 {27, $[OC(CH_3)_3]^+$ } and 57 {100, $[C(CH_3)_3]^+$.

N-tert-Butoxycarbonyl-*N*'-[(2*S*)-*N*-chloroacetylprolyl]-*N*'-methylhydrazine 20

To a solution of (2S)-N-chloroacetylproline 19 (1.92 g, 10 mmol) in dry THF (25 cm³) was added N-methylmorpholine (1.12 cm³, 10 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (1.36 cm³, 10 mmol) was added with stirring and the solution was stirred at -15 °C for 2 min. A solution of *N-tert*-butoxycarbonyl-*N*⁻methylhydrazine 18 (1.46 g, 10 mmol) in dry THF (20 cm³) was then added. The reaction mixture was allowed to warm to room temperature and then stirred for 6 h. The hydrochloride salts were filtered off and the solvents removed under reduced pressure. The oily residue was dissolved in CH_2Cl_2 (25 cm³) and washed with 0.5 mol dm⁻³ HCl $(2 \times 15 \text{ cm}^3)$ and 5% aqueous sodium carbonate $(2 \times 15 \text{ cm}^3)$ cm³). The organic phase was then dried (MgSO₄) and the solvent removed under reduced pressure to give a yellow oil. The crude material was purified by silica column chromatography using ethyl acetate as the eluent to give the product as a white solid (2.17 g, 68%), mp 64–67 °C (HRMS: found $[M + H]^+$, 320.1373. $C_{13}H_{23}^{-35}ClN_{3}O_{4}$ requires 320.1376); $[a]_{D}^{22}$ -11.0 (c 0.5 in MeOH); v_{max}(Nujol)/cm⁻¹ 3227 (NH), 1730 (carbamate CO), 1695 (methylamide CO), 1678 (tertiary amide CO) and 771 (C-Cl); δ_H(200 MHz; C²HCl₃) 1.48 [9 H, s, C(CH₃)₃], 1.74-2.45 (4 H, m, γCH₂ and βCH₂), 3.11 (3 H, s, NCH₃), 3.63 (2 H, m, δCH₂), 4.04 (1 H, d, J 13.0, ¹/₂COCH₂), 4.13 (1 H, d, J 13.0, $\frac{1}{2}$ COCH₂), 4.97 (1 H, dd, J_1 5.6, J_2 6.2, α CH) and 7.90 (1 H, s, NH); $\delta_{\rm C}(50.31 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 25.15 (γ CH₂), 28.47 [C(CH₃)₃], 28.55 (βCH₂), 35.54 (NCH₃), 42.48 (COCH₂), 47.59 (δCH₂), 57.32 (αCH), 81.49 [C(CH₃)₃], 154.96 (CONH), 165.31

 $\begin{array}{l} ({\it CONCH_3}) \mbox{ and } 173.73 \ ({\it COCH_2}); \ {\it m/z} \ (CI) \ 320 \ (3\%, [M + H]^+), \\ 284 \ (100, \ [M - Cl]^+), \ 240 \ \{91, \ [M - Cl - C(CH_3)_3 + H]^+\}, \\ 225 \ \{9, \ [M - Cl - OC(CH_3)_3 + 2H^+\}, \ 170 \ \{15, \ [M - CH_2Cl - CO_2C(CH_3)_3 + H]^+\} \ \ and \ 155 \ \{4, \ [M - CH_2Cl - NHCO_2C(CH_3)_3 + H]^+\}. \end{array}$

(9a.5)-2-Methyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo[2,1-*d*]-[1,2,5]triazepine-1,5-dione 17

Hydrogen chloride gas was bubbled through a solution of Ntert-butoxycarbonyl-N'-[(2S)-N-chloroacetylprolyl]-N'-methylhydrazine 20 (1.60 g, 5 mmol) in ethyl acetate (30 cm³) for 20 min at 0 °C. The solvent was then removed under reduced pressure and the resultant hygroscopic white solid was dissolved in water (25 cm³) and NaOH solution (1 mol dm⁻³) added until the solution reached pH 9. The solution was then extracted with ethyl acetate $(3 \times 20 \text{ cm}^3)$, the combined organic fractions were dried (MgSO₄) and the solvent removed under reduced pressure to give the product as a white solid (0.59 g, 64%), mp 75-79 °C (HRMS: found M⁺, 183.1001. C₈H₁₃N₃O₂ requires 183.1007); [a]²²_D -40.2 (c1.0 in MeOH); v_{max}(Nujol)/cm⁻¹ 3440 (NH), 1652 (methylamide CO) and 1637 (tertiary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.82–2.38 (A and B, 4 H, m, γ CH₂ and β CH₂), 3.18 (A, 3 H, s, NCH₃), 3.22 (B, 3 H, s, NCH₃), 3.69 (A and B, 2 H, m, δCH₂), 3.91 (A and B, 1 H, s, NH), 4.03 (A and B, 1 H, d, J12.4, ¹/₂COCH₂), 4.13 (A and B, 1 H, d, J 12.4, ¹/₂COCH₂), 5.39 (B, 1 H, dd, J₁ 3.2, J₂ 8.6, αCH) and 5.47 (A, 1 H, dd, J₁ 3.9, J₂ 7.9, α CH); δ_{c} (50.31 MHz; C²HCl₃) 22.22 (*B*, γ CH₂), 24.52 (*A*, γCH₂), 29.07 (A, βCH₂), 31.67 (B, βCH₂), 38.26 (A, NCH₃), 38.43 (B, NCH₃), 42.41 (A and B, COCH₃), 47.44 (A, \deltaCH₃), 47.65 (B, \deltaCH₂), 57.46 (A, aCH), 57.88 (B, aCH), 164.51 (A, CONCH₃), 165.13 (B, CONCH₃) and 172.97 (A and B, $COCH_2$; m/z (EI) 183 (45%, M⁺), 155 (15, [M -NCH₃ + H]⁺), 139 (65, [M - NHNCH₃]⁺), 125 (20, [M - CH₂-NHNCH₃]⁺), 111 (40, [M - CONCH₃NH]⁺) and 70 (100, $[C_4H_8N]^+$).

(2S,2'S)-N-(2'-Chloropropionyl)proline methyl ester 22a

To a solution of N-methylmorpholine (1.12 cm³, 10 mmol) in dry THF (20 cm³) was added (2S)-2-chloropropionic acid 21a,²⁴ (1.09 g, 10 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (1.36 cm³, 10 mmol) was added with stirring and the resultant suspension was stirred at -15 °C for 2 min. A mixture of (2S)-proline methyl ester hydrochloride (1.66 g, 10 mmol) and N-methylmorpholine (1.12 cm³, 10 mmol) in dry DMF (5 cm³) was added. The reaction mixture was allowed to warm to room temperature and then stirred for 3 h. The hydrochloride salts were filtered off and the solvents removed under reduced pressure. The resulting clear oil was dissolved in CH₂Cl₂ (25 cm³) and washed with 0.5 mol dm⁻³ HCl (2×15 cm³) and 5% aqueous sodium carbonate $(2 \times 15 \text{ cm}^3)$. The organic phase was then dried (MgSO₄) and the solvent removed under reduced pressure to yield a yellow oil which was purified by silica chromatography using ethyl acetate-light petroleum (1:1) as the eluent to give the product as a clear oil (1.56 g, 71%) (HRMS: found [M + H]⁺, 220.0747. C₉H₁₅³⁵ClNO₃ requires 220.0742); $[a]_{D}^{22}$ -70.8 (c 1.0 in MeOH); v_{max} (thin film)/cm⁻¹ 1746 (ester CO), 1659 (amide CO) and 1433 (C–O); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.62 (3 H, d, J6.8, CHCH₃), 1.85-2.32 (4 H, m, γCH_2 and $\beta CH_2),~3.51{-}3.92$ (2 H, m, $\delta CH_2),~3.69$ (3 H, s, OCH₃), 4.47 (1 H, dd, J₁ 3.4, J₂ 6.6, aCH) and 4.49 (1 H, q, J 6.8, CHCH₃); δ_{c} (50.31 MHz; C²HCl₃) 20.50 (t, CHCH₃), 21.32 (c, CHCH₃), 21.99 (c, γCH₂), 24.79 (t, γCH₂), 29.05 (t, βCH₂), 31.35 (c, βCH₂), 46.91 (t and c, δCH₂), 51.17 (t and c, CHCH₃), 52.13 (t, OCH₃), 52.48 (c, OCH₃), 59.18 (t, αCH), 59.28 (c, aCH), 167.52 (t and c, COCHCH₃), 172.07 (c, CO₂CH₃) and 172.23 (t, CO_2CH_3); m/z (EI) 220 (7%, $[M + H]^+$), 184 (5, $[M - Cl]^+$, 170 (54, $[M - Cl - CH_3 + H]^+$), 160 (84, $[M - CO_2CH_3]^+)$, 128 (58, $[M - Cl - CO_2CH_3 + 3H]^+$) and 70 (100, $[C_4H_8N]^+$).

(2S,2' R)-N-(2'-Chloropropionyl)proline methyl ester 22b

This compound was prepared in a manner identical with that for the (2S, 2'S) diastereomer **22a**, using (2R)-2-chloropropionic acid **21b**²⁹ (1.09 g, 10 mmol) to give the product as colourless crystals (1.82 g, 83%), mp 118-120 °C (Found: C, 49.2; H, 6.4; N, 6.4. C₉H₁₄ClNO₃ requires C, 49.05; H, 6.6; N, 6.4%); [a]_D²² -113.9 (c 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 1751 (ester CO), 1656 (amide CO) and 1454 (C–O); $\delta_{\rm H}(200~{\rm MHz};~{\rm C^2HCl_3})$ 1.64 (c, 3 H, d, J6.4, CHCH₃), 1.66 (t, 3 H, d, J6.6, CHCH₃), 1.80-2.33 (t and c, 4 H, m, γCH₂ and βCH₂), 3.46-3.70 (t, 2 H, m, δCH₂), 3.74 (t, 3 H, s, OCH₃), 3.77 (c, 3 H, s, OCH₃), 3.82-3.98 (c, 2 H, m, δCH_2), 4.24 (c, 1 H, q, J 6.4, CHCH₃), 4.46 $(t, 1 \text{ H}, \text{ dd}, J_1 3.2, J_2 8.4, \alpha \text{CH})$ and $4.71 (c, 1 \text{ H}, \text{ dd}, J_1 5.0, J_2$ 5.6, α CH); δ_{C} (74.76 MHz; C²HCl₃) 20.82 (*t*, CH*C*H₃), 20.94 (c, CHCH₃), 22.78 (c, γCH₂), 25.24 (t, γCH₂), 29.41 (t, βCH₂), 31.44 (c, βCH₂), 47.26 (c, δCH₂), 47.35 (t, δCH₂), 51.29 (t, CHCH₃), 51.39 (c, CHCH₃), 52.62 (t, OCH₃), 53.27 (c, OCH₃), 59.47 (c, aCH), 59.62 (t, aCH), 167.99 (t, COCHCH₃), 168.35 (c, COCHCH₃), 172.48 (t, CO₂CH₃) and 172.78 (c, CO_2CH_3 ; m/z (EI) 220 (3%, $[M + H]^+$), 184 (4, $[M - Cl]^+$), 170 $(5, [M - Cl - CH_3 + H]^+), 160 (63, [M - CO_2CH_3]^+), 127 (58,$ $[M - Cl - CO_2CH_3 + 2H]^+$) and 70 (100, $[C_4H_8N]^+$).

(2S,2'S)-N-(2'-Chloropropionyl)proline 23a

To a solution of (2S,2'S)-N-(2'-chloropropionyl)proline methyl ester 22a (2.20 g, 10 mmol) in methanol (20 cm³) was added 1 mol dm⁻³ aqueous sodium hydroxide (22 cm³). The solution was stirred at room temperature for 1 h and then 1 mol dm⁻³ aqueous HCl (10 cm³) was added. Methanol was removed under reduced pressure and a second portion of 1 mol dm⁻³ aqueous HCl (10 cm³) was added. The precipitated white solid was filtered and recrystallised from methanol to give the product as colourless crystals (1.77 g, 84%), mp 164–165 °C (Found: C, 46.95; H, 5.95; N, 6.75. $C_8H_{12}CINO_3$ requires C, 46.75; H, 5.9; N, 6.8%); $[a]_{D}^{22}$ -62.1 (*c* 1.0 in MeOH); ν_{max} (Nujol)/cm⁻¹ 3045 (OH), 1739 (acid CO) and 1624 (amide CO); $\delta_{\rm H}(200 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 1.68 (3 H, d, J 6.6, CH₃), 1.92– 2.33 (4 H, m, βCH₂ and γCH₂), 3.64 (1 H, m, ½δCH₂), 3.78 (1 H, m, $\frac{1}{2}\delta$ CH₂), 4.53 (1 H, q, J 6.6, CHCH₃), 4.57 (1 H, dd, $J_1 = J_2$ 5.8, α CH) and 9.19 (1 H, s, CO₂H); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 20.99 (CH₃), 25.34 (γCH₂), 29.39 (βCH₂), 47.79 (δCH₂), 51.67 (CHCH₃), 60.11 (aCH), 169.23 (NCO) and 175.57 (CO₂H); m/z (EI) 205 (14%, M^+), 172 (46, $[M - Cl + 2H]^+$), 144 (66, $[M - Cl + 2H]^+$) CHClCH₃ + 2H]⁺), 116 (59, [M - COCHClCH₃ + 2H]⁺), 97 (87, [C₄H₈NCO]⁺) and 70 (100, [C₄H₈N]⁺).

(2*S*,2'*R*)-*N*-(2'-Chloropropionyl)proline 23b

This was prepared in a manner identical with that for the (2S,2'S) diastereomer using (2S,2'R)-N-(2'-chloropropionyl)proline methyl ester **22b**, except that after the addition of the second portion of 1 mol dm^{-3} aqueous HCl (10 cm³), the solution was extracted with ethyl acetate $(3 \times 50 \text{ cm}^3)$. The combined organic fractions were dried (MgSO₄) and the solvent removed under reduced pressure to give the product as a white solid (1.54 g, 79%), mp 102-106 °C (HRMS: found $[M + H]^+$, 206.0589. $C_8 H_{13}^{35} ClNO_3$ requires 206.0584); $[a]_D^{22}$ -110.2 (c 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3344 (OH), 1750 (acid CO) and 1616 (amide CO); $\delta_{\rm H}(200~{\rm MHz};~{\rm C^2HCl_3})$ 1.64 (c, 3 H, d, J 6.4, CH₃), 1.65 (t, 3 H, d, J 6.6, CH₃), 1.85-2.43 (t and c, 4 H, m, βCH_2 and γCH_2), 3.58 (t and c, 1 H, m, ¹/₂δCH₂), 3.89 (t and c, 1 H, m, ¹/₂δCH₂), 4.32 (c, 1 H, q, J 6.4, CHCH₃), 4.48 (t, 1 H, q, J 6.6, CHCH₃), 4.52 (t, 1 H, dd, J₁ 4.3, J₂ 6.7, αCH), 4.72 (c, 1 H, dd, J₁ 4.3, J₂ 6.7, αCH) and 10.25 (t and c, 1 H, s, CO₂H); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 20.45 (t and c, CH₃), 22.37 (c, γCH₂), 24.70 (t, γCH₂), 28.88 (t, βCH₂), 30.99 (c, βCH₂), 47.21 (c, δCH₂), 47.26 (t, δCH₂), 51.07 (t and c, CHCH₃), 59.36 (t and c, αCH), 168.53 (t, NCO), 168.81 (c, NCO), 173.92 (c, CO₂H) and 174.73 (t, $CO_{2}H$); m/z (CI) 206 (58%, [M + H]⁺), 170 (100, [M - $Cl]^+$), 156 (23, $[M - Cl - CH_3 + H]^+$), 142 (6, $[M - Cl - CH_3 + H]^+$)

 $\rm CHCH_3]^+), \ 113 \ (12, \ [M-CHCl-CO_2]^+) \ and \ 97 \ (3, \ [C_4H_{8^-}NCO]^+).$

N-tert-Butoxycarbonyl-*N*-[(2*S*,2'*S*)-*N*-(2'-chloropropionyl)prolyl]-*N*-methylhydrazine 24a

To a solution of (2S, 2'S) - N - (2' - chloropropionyl) proline 23a (2.06 g, 10 mmol) in dry THF (40 cm³) was added *N*-methylmorpholine (1.12 cm³, 10 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (1.36 cm³, 10 mmol) was added with stirring and the resulting suspension stirred at -15 °C for 2 min. A solution of N-tert-butoxycarbonyl-Nmethylhydrazine 18 (1.46 g, 10 mmol) in dry THF (10 cm³) was added and the reaction mixture was allowed to warm to room temperature and left to stir for 12 h. The hydrochloride salts were filtered off and the solvent removed under reduced pressure. The oily residue was dissolved in CH₂Cl₂ (25 cm³) and washed with 0.5 mol dm⁻³ HCl (2×15 cm³) and 5% aqueous sodium carbonate $(2 \times 15 \text{ cm}^3)$. The organic phase was then dried (MgSO₄) and the solvent removed under reduced pressure to yield a yellow oil which was purified by silica chromatography using ethyl acetate-light petroleum (1:1) as the eluent to give the product as a white solid (2.31 g, 69%), mp 81-83 °C (Found: C, 50.6; H, 7.25; N, 12.5. C₁₄H₂₄ClN₃O₄ requires C, 50.4; H, 7.25; N, 12.6%) (HRMS: found [M + H]⁺, 334.1538. $C_{14}H_{25}^{35}ClN_{3}O_{4}$ requires 334.1536); $[a]_{D}^{22} - 16.7$ (c1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3327 (NH), 1735 (carbamate CO), 1685 (methylamide CO) and 1646 (tertiary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.48 [9 H, s, C(CH₃)₃], 1.64 (3 H, d, J 6.6, CHCH₃), 1.82-2.31 (4 H, m, βCH₂ and γCH₂), 3.12 (3 H, s, NCH₃), 3.52-3.89 (2 H, m, δCH₂), 4.53 (1 H, q, J6.6, CHCH₃), 4.98 (1 H, dd, $J_1 = J_2$ 6.3, α CH) and 7.87 (1 H, s, NH); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 21.10 (CHCH₃), 25.20 (γCH₂), 28.64 [C(CH₃)₃], 28.78 (βCH₂), 35.55 (NCH₃), 47.77 (δCH₂), 51.68 (CHCH₃), 57.77 (αCH), 81.78 [C(CH₃)₃], 155.11 (CO₂Bu⁴), 168.09 (CONCH₃) and 173.86 (COCHCH₃); m/z (CI) 334 (33%, $[M + H]^+$), 300 (15, $[M - Cl + 2H]^+$), 278 {100, $[M - C(CH_3)_3 + 2H]^+$ }, 244 {28, $[M - Cl - C(CH_3)_3 + 3H]^+$, 234 {12, $[M - CO_2C(CH_3)_3 + 3H]^+$ $2H]^+$ and 200 {13, [M - Cl - CO₂C(CH₃)₃ + 3H]⁺}.

N-tert-Butoxycarbonyl-N'-[(2S,2'R)-N-(2'-chloropropionyl)-prolyl]-N'-methylhydrazine 24b

This compound was prepared in a manner identical with that for the (2S,2'S) diastereomer **24a**, using (2S,2'R)-N-(2'chloropropionyl)proline 23b (2.06 g, 10 mmol) to give the product as a white solid (2.47 g, 74%), mp 132-135 °C (Found: C, 50.6; H, 7.3; N, 12.5. C₁₄H₂₄ClN₃O₄ requires C, 50.4; H, 7.25; N, 12.6%) (HRMS: found $[M + H]^+$, 334.1539. $C_{14}H_{25}^{35}ClN_3O_4$ requires 334.1536); $[a]_{D}^{22}$ -20.3 (c 0.2 in MeOH); v_{max} (Nujol)/ cm⁻¹ 3209 (NH),1728 (carbamate CO), 1687 (methylamide CO) and 1652 (tertiary amide CO); $\delta_{\rm H}(200 \text{ MHz}; \text{ C}^2\text{HCl}_3)$ 1.47 [t, 9 H, s, C(CH₃)₃], 1.51 [c, 9 H, s, C(CH₃)₃], 1.60 (c, 3 H, d, J 6.6, CHCH₃), 1.65-2.48 (t and c, 4 H, m, γCH₂ and βCH₂), 1.65 (t, 3 H, d, J6.6, CHCH₃), 3.11 (t, 3 H, s, NCH₃), 3.13 (c, 3 H, s, NCH₃), 3.57 (*t* and *c*, 1 H, m, $\frac{1}{2}\delta$ CH₂), 3.89 (*t* and *c*, 1 H, m, ¹/₂δCH₂), 4.49 (*t* and *c*, 1 H, q, *J* 6.8, C*H*CH₃), 4.96 (*t* and *c*, 1 H, dd, J_1 4.2, J_2 7.8, α CH) and 7.92 (*t* and *c*, 1 H, s, NH); δ_c (50.31) MHz; C²HCl₃) 20.75 (c, CHCH₃), 21.05 (t, CHCH₃), 22.37 (c, γCH₂), 24.71 (t, γCH₂), 28.03 [c, C(CH₃)₃], 28.39 [t, C(CH₃)₃], 31.37 (c, βCH₂), 35.17 (t, NCH₃), 36.05 (c, NCH₃), 47.37 (c, δCH₂), 47.50 (t, δCH₂), 51.15 (t, CHCH₃), 53.86 (c, CHCH₃), 57.25 (t, αCH), 58.02 (c, αCH), 81.38 [t and c, C(CH₃)₃], 155.17 (t and c, CO₂Bu⁴), 167.95 (t and c, CONCH₃) and 173.86 (t and *c*, *C*OCHCH₃); m/z (CI) 334 (41%, [M + H]⁺), 300 (11, $[M - Cl + 2H]^+$), 278 {100, $[M - C(CH_3)_3 + 2H]^+$ }, 234 {12, $[M - CO_2C(CH_3)_3 + 2H]^+$, 219 {11, $[M - NHCO_2C(CH_3)_3 + 2H]^+$ $2H]^+$ and 185 {8, [M - Cl - CH₃ - CO₂C(CH₃)₃ + 3H]⁺}.

(4*R*,9a*S*)-2,4-Dimethyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 17a

Hydrogen chloride gas was bubbled through a solution of N-tert-butoxycarbonyl-N-[(2S,2'S)-N-(2'-chloropropionyl)-

prolyl]-N-methylhydrazine 24a (1.35 g, 5 mmol) in ethyl acetate (30 cm³) for 20 min at 0 °C. The solvent was then removed under reduced pressure and the resulting hygroscopic white solid dissolved in methanol (25 cm³) and \overline{N} -methylmorpholine (1.12 cm³, 10 mmol) was added. The solvent was then removed under reduced pressure and the remaining residue dissolved in water and extracted with ethyl acetate $(3 \times 30 \text{ cm}^3)$. The combined organic fractions were then dried (MgSO₄) and the solvent removed under reduced pressure to yield an off-white solid which was purified by silica chromatography using ethyl acetate-methanol (95:5) as the eluent to give the product as a white solid (0.61 g, 62%), mp 108–110 °C (HRMS: found M^+ , 197.1159. $C_9H_{15}N_3O_2$ requires 197.1163); $[a]_D^{22} - 82.6$ (c 0.5 in MeOH); v_{max}(Nujol)/cm⁻¹ 3336 and 3215 (NH), 1655 (tertiary amide CO) and 1641 (secondary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.66 (3 H, d, J6.6, CHCH₃), 1.83-2.36 (4 H, m, γCH₂ and \(\beta\)CH2), 3.18 (3 H, s, NCH3), 3.76 (2 H, m, \delta\)CH2), 4.54 (1 H, q, J 6.6, $CHCH_3$) and 5.48 (1 H, dd, J_1 4.6, J_2 8.0, αCH); $\delta_{\rm C}(50.31 \text{ MHz}; \text{ C}^2\text{HCl}_3) 21.14 (\text{CH}_{\rm CH}_3), 25.28 (\gamma\text{CH}_2), 29.69$ (βCH₂), 39.25 (NCH₃), 47.89 (δCH₂), 51.97 (CHCH₃), 57.54 (aCH), 167.80 (CONCH₃) and 174.08 (COCHCH₃); m/z (EI) 197 (15%, M^+), 182 (9, $[M - CH_3]^+$), 153 (17, $[M - CH_3 - CH_3]^+$) NCH_{3}^{+}), 139 (5, $[M - NHNCH_{3} - CH_{3} + H]^{+}$), 127 (15, $[M - CHCH_3NHNCH_3 + 2H]^+$) and 70 (100, $[C_4H_8N]^+$).

(4*S*,9a*S*)-2,4-Dimethyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo-[2,1-*d*][1,2,5]triazepine-1,5-dione 17b

This compound was prepared in a manner identical with that for the (4R,9aS) diastereomer 17a, starting from N-tertbutoxycarbonyl-N'-[(2S,2'R)-N-(2'-chloropropionyl)prolyl]-N-methylhydrazine 24b (1.35 g, 5 mmol) instead of N-tertbutoxycarbonyl-N'-[(2S,2'S)-N-(2'-chloropropionyl)prolyl]-N-methylhydrazine 24a to give the product as a white solid (0.63 g, 64%), mp 112–115 $^\circ\!C$ (HRMS: found M⁺, 197.1159. $C_9H_{15}N_3O_2$ requires 197.1163); $[a]_D^{22} - 119.4$ (*c* 1.0 in MeOH); ν_{max} (Nujol)/cm⁻¹ 3299 and 3210 (NH), 1652 (tertiary amide CO) and 1634 (secondary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.57 (B, 3 H, d, J 6.6, CHCH₃), 1.62 (A, 3 H, d, J 6.8, CHCH₃), 1.78–2.34 (A and B, 4 H, m, γCH₂ and βCH₂), 3.14 (A, 3 H, s, NCH₃), 3.18 (B, 3 H, s, NCH₃), 3.59 (A and B, 1 H, m, ¹/₂δCH₂), 3.92 (A and B, 1 H, m, ¹/₂δCH₂), 4.48 (A and B, 1 H, J 6.7, CHCH₃), 5.40 (A, 1 H, dd, J₁ 4.4, J₂ 7.8, αCH) and 5.42 (B, 1 H, dd, J_1 4.4, J_2 8.6, α CH); $\delta_{\rm C}$ (50.31 MHz; C²HCl₃) 21.03 (A, CHCH₃), 21.14 (B, CHCH₃), 22.69 (B, γCH₂), 25.03 (A, γCH₂), 29.42 (A, \(\beta\)CH2), 32.09 (B, \(\beta\)CH2), 38.69 (A, NCH3), 39.21 (B, NCH₃), 47.84 (A and B, 8CH₂), 51.62 (A, CHCH₃), 51.71 (B, CHCH₃), 57.40 (A, aCH), 58.40 (B, aCH), 167.49 (A, COCHCH₃), 168.33 (B, COCHCH₃), 173.45 (A, CONCH₃) and 173.92 (B, CONCH₃); m/z (EI) 197 (5%, M⁺), 182 (19, $[M - CH_3]^+$), 153 (7, $[M - CH_3 - NCH_3]^+$), 139 (6, $[M - CH_3 - NCH_3]^+$) NHNCH₃ - CH₃ + H]⁺), 127 (23, [M - CHCH₃NHNCH₃ + 2H]⁺), 97 (37, [C₄H₈NCO]⁺) and 70 (100, [C₄H₈N]⁺).

(4R,9aS)-2,4-Dimethyl-3-[(2S)-N-benzyloxycarbonylphenylalanyl]-2,3,4,5,7,8,9,9a-octahydro-1H-pyrrolo[2,1-d][1,2,5]-triazepine-1,5-dione 17c

This compound was prepared in a manner identical with that for *N-tert*-butoxycarbonyl-*N*⁻[(2*S*)-*N*-chloroacetylprolyl]-*N*⁻ methylhydrazine **20**, using (2*S*)-*N*-benzyloxycarbonylphenylalanine (1.50 g, 5 mmol) instead of (2*S*)-*N*-chloroacetylproline **19** and (4*R*,9a*S*)-2,4-dimethyl-1,2,5-triazepine-1,5-dione **17a** (1.99 g, 5 mmol) instead of *N-tert*-butoxycarbonyl-*N*⁻ methylhydrazine **18**. The crude material was purified by silica column chromatography using ethyl acetate–light petroleum (7:3) as the eluent to give the product as a white solid (1.96 g, 82%), mp 82–85 °C (HRMS: found [M + H]⁺, 479.2282. C₂₆H₃₁N₄O₅ requires 479.2294); [a]₂²² +0.8 (*c* 1.0 in MeOH); v_{max} (Nujol)/cm⁻¹ 3270 (NH), 1717 (carbamate CO), 1695 (methylamide CO), 1659 (secondary amide CO) and 1645 (tertiary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.59 (3 H, d, *J* 6.6, CHCH₃), 1.68-2.35 (4 H, m, YCH₂ and BCH₂), 2.89 (3 H, s, NCH₃), 3.09 (2 H, d, J 7.6, ¹/₂CHCH₂Ph), 3.14 (2 H, d, J 7.0, $\frac{1}{2}$ CHCH₂Ph), 3.56 (2 H, m, δ CH₂), 4.44 (1 H, dd, J₁ 7.4, J₂ 8.8, αCH), 4.47 (1 H, q, J6.6, CHCH₃), 4.70 (1 H, m, CHNH), 5.08 (2 H, d, J2.2, OCH₂Ph), 5.51 (1 H, d, J7.2, NH), 7.22 (5 H, m, Ar–H, CHCH₂Ph) and 7.32 (5 H, s, Ar–H, OCH₂Ph); $\delta_{\rm C}(50.31$ MHz; C²HCl₃) 21.15 (CHCH₃), 25.37 (γCH₂), 29.21 (βCH₂), 35.66 (CHCH₂Ph), 38.66 (NCH₃), 47.90 (δCH₂), 51.99 (CHCH₃), 55.46 (CHNH), 57.13 (αCH), 67.60 (OCH₂Ph), 127.75 (Ar-CH para), 128.51 (Ar-CH ortho), 128.72 (Ar-CH para), 129.07 (Ar-CH ortho), 129.30 (Ar-CH meta), 129.96 (Ar–CH *ortho*), 136.37 (Ar–C quaternary), 136.72 (Ar–C quaternary), 156.53 (CONH), 168.26 (CONCH₃), 170.90 (COCHCH₃) and 173.65 (COCHCH₂Ph); m/z (CI) 513 (100%, $[M+2NH_{3}+H]^{\scriptscriptstyle +}), \ \ 479 \ \ (19, \ \ [M+H]^{\scriptscriptstyle +}), \ \ 403 \ \ (3, \ \ [M-1]{}$ $OCH_2Ph - Ph + H]^+$), 254 (5, $[M - NHCO_2CH_2Ph - Ph +$ 3H]⁺) and 181 (32, [M - COCHCH₂Ph - NHCO₂CH₂Ph - $CH_{3}]^{+}).$

N-tert-Butoxycarbonyl-N -phenylhydrazine 27

To a solution of phenylhydrazine (5.0 cm³, 30.8 mmol) in diethyl ether (125 cm³) was added di-tert-butyl dicarbonate (11.08 g, 50.8 mmol), and the mixture was stirred at room temperature for 3 days. The solvent was then removed under reduced pressure to yield a yellow solid which was washed with light petroleum (25 cm³), 0.5 mol dm⁻³ hydrochloric acid (25 cm³) and water (25 cm³). The resulting N-protected N'-phenylhydrazine 27 was dried under reduced pressure and then recrystallised from ethyl acetate-light petroleum to give colourless needles (9.14 g, 86%), mp 88-90 °C (found: C, 63.5; H, 7.8; N, 13.65. C₁₁H₁₆N₂O₂ requires C, 63.4; H, 7.7; N, 13.45%); v_{max}-(thin film)/cm⁻¹ 3354 (amine NH), 3278 (amide NH) and 1700 (CO); δ_H(200 MHz; C²HCl₃) 1.48 [9 H, s, C(CH₃)₃], 5.90 (1 H, s, NHPh), 6.64 (1 H, s, NHBoc), 6.80 (2 H, d, J8.4, Ar-H ortho), 6.89 (1 H, t, J7.5, Ar-H para) and 7.23 (2 H, t, J 8.0, Ar-H *meta*); $\delta_{\rm C}(74.76 \text{ MHz}; C^2 \text{HCl}_3)$ 28.15 [C(CH₃)₃], 81.03 [C(CH₃)₃], 112.88 (Ar-CH ortho), 120.56 (Ar-CH para), 129.00 (Ar-CH meta), 148.33 (Ar-C quaternary) and 156.25 $(CO_2Bu'); m/z$ (EI) 208 (9%, M⁺), 152 (81, $[M - C_4H_9 + H]^+),$ 108 (57, $[M - CO_2C_4H_9 + H]^+$), 92 (44, $C_6H_5NH^+$), 77 (52, $C_6H_5^+$), 65 (31, $C_5H_5^+$) and 57 (100, $C_4H_9^+$).

N-(tert-Butoxycarbonyl)-*N*-[(2*S*)-*N*-chloroacetylprolyl]-*N*-phenylhydrazine 28

To a stirred solution of (2S)-N-chloroacetylproline 19 (0.33 g, 1.7 mmol) and pyridine (0.19 cm³, 2.4 mmol) in dichloromethane (10 cm³) was added, dropwise, thionyl chloride (0.15 cm³, 2.0 mmol). After 10 min, a solution of hydrazide 27 (0.30 g, 1.4 mmol) and DMAP (0.39 g, 32 mmol) in dichloromethane $(10\ {\rm cm^3})$ was added in one portion. After 3–4 days, the orange solution was washed with 0.1 mol dm⁻³ hydrochloric acid $(2 \times 20 \text{ cm}^3)$, 5% aqueous sodium hydrogen carbonate $(2 \times 20 \text{ cm}^3)$ cm³) and brine (20 cm³). The organic phase was dried (MgSO₄) and the solvent removed under reduced pressure to yield a brown oil which was purified by silica chromatography using light petroleum-ethyl acetate (3:2) as eluent to give the desired compound as a pale brown oil (0.34 g, 62%) (HRMS: found $[M - OC_4H_9]^+$, 308.0808. $C_{14}H_{15}CIN_3O_3$ requires 308.0802); $[a]_D^{22} - 21.5$ (*c* 0.4 in MeOH); v_{max} (thin film)/cm⁻¹ 1733 (ester CO), 1694 (secondary amide CO), 1654 (tertiary amide CO) and 1162 (ester C-O); $\delta_{\rm H}$ (200 MHz; C²HCl₃), 1.38 [9 H, s, C(CH₃)₃], 1.76-2.12 (4 H, m, βCH₂ and γCH₂), 3.38-3.63 (2 H, m, δCH₂), 4.00-4.09 (2 H, m, CH₂Cl), 4.40 (c, 1 H, m, αCH), 5.15 (t, 1 H, m, aCH), 7.06-7.52 (5 H, m, Ar-H), 9.05 (c, 1 H, br s, NH) and 9.35 (t, 1 H, br s, NH); $\delta_{\rm C}$ (74.76 MHz; C²HCl₃) 25.28 (c, \(\gamma\)CH2), 25.42 (t, \(\gamma\)CH2), 28.60 [CO(CH3)3], 28.95 (t, βCH₂), 29.20 (c, βCH₂), 42.50 (CH₂Cl), 47.86 (δCH₂), 58.17 (t, αCH), 58.39 (c, αCH), 81.86 [C(CH₃)₃], 125.15, 126.95 (Ar-CH ortho), 128.37, 128.87 (Ar-CH para), 129.11, 129.95 (Ar-CH *meta*), 141.56, 142.14 (Ar–C quaternary), 154.97 (*t*, CO₂Bu⁴), 155.96 (*c*, *CO*₂Bu⁴), 165.39 (*t*, *CO*CH₂Cl), 165.83 (*c*, *CO*CH₂Cl) and 173.47 (*C*ONPh); *m/z* (EI) 382 (2%, M⁺), 308 (6, $[M - OC_4H_9]^+$), 281 (14, $[M - CO_2C_4H_9]^+$), 174 (55, $[C_7H_9-NO_2Cl + H]^+$), 146 (100, $C_6H_9NOCl^+$), 70 (77, $[C_4H_8N + H]^+$) and 57 (57, $C_4H_9^+$).

(9a.*S*)-2-Phenyl-2,3,4,5,7,8,9,9a-octahydro-1*H*-pyrrolo[2,1-*d*]-[1,2,5]triazepine-1,5-dione 26

Dry hydrogen chloride gas was bubbled through a solution of the dipeptide **28** (0.84 g, 2.2 mmol) in ethyl acetate (25 cm³) for 40 min. The solvent was removed under reduced pressure and the resulting yellow solid was redissolved in hydrochloric acid $(0.5 \text{ mol } \text{dm}^{-3}, 25 \text{ cm}^3)$. The solution was washed with diethyl ether (30 cm³), and basified with aqueous sodium hydroxide (1 mol dm⁻³, 35 cm³). The solution was allowed to stand for 30 min and then extracted with ethyl acetate $(3 \times 50 \text{ cm}^3)$. The organic phase was washed with water (50 cm³), dried (MgSO₄) and the solvent removed under reduced pressure to yield the compound 26 as an orange-brown oil (0.36 g, 68%) (HRMS: found M⁺, 245.1170. $C_{13}H_{15}N_3O_2$ requires 245.1164); v_{max} (thin film)/cm $^{-1}$ 3336 (NH) and 1650 (br, CO); $\delta_{\rm H}(\rm 300~MHz;$ C²HCl₃), 1.84–1.97 (3 H, m, ¹₂βCH₂ and γCH₂), 2.08–2.17 (1 H, m, ¹₂βCH₂), 3.51-3.67 (2 H, m, δCH₂), 3.95-4.06 (2 H, m, $CH_{5}NH$, 4.41 (1 H, m, αCH), 4.81 (1 H, br s, NH) and 7.20– 7.50 (5 H, m, Ar–H); $\delta_{\rm C}$ (74.76 MHz; C²HCl₃) 25.55 (γ CH₂), 29.84 (βCH₂), 42.65 (CH₂NH), 48.04 (δCH₂), 58.17 (αCH), 127.99 (Ar-CH ortho), 128.93 (Ar-CH para), 130.08 (Ar-CH meta), 142.35 (Ar-C quaternary), 165.28 (COCH₂N) and 171.15 (CONPh); m/z (EI) 245 (15%, M⁺), 217 (33, [M -CO]⁺), 110 (40, C₆H₈NO⁺), 93 (57, C₆H₅NH₂⁺), 77 (42, C₆H₅⁺) and 70 (100, C₄H₈N⁺).

(2S)-2-Benzylhydantoic acid 32

To a stirred suspension of (2S)-phenylalanine 31 (5.00 g, 30.3 mmol) in water (80 cm³) was added potassium cyanate (22.9 g, 0.28 mol), and the resulting mixture was heated at 60 °C for 4 h. The clear solution was ice-cooled and carefully acidified with concentrated hydrochloric acid (37%, 50 cm³). The precipitate thus formed was filtered off, washed with cold water and dried under vacuum to yield (2S)-2-benzylhydantoic acid (5.40 g, 86%), mp 189–190 °C (lit.,²³ 189 °C); $[a_{D}^{22} + 38.5 (c 1.0 \text{ in MeOH})$ [lit.,²³ 45.0 (c 1.0 in EtOH)]; v_{max} (Nujol)/cm⁻¹ 3450 (N–H), 3297 (N-H), 2926 (OH), 1693 (acid CO) and 1559 (amide CO); δ_H(200 MHz; [²H₆]DMSO) 2.82-3.08 (2 H, m, CH₂), 4.35 (1 H, m, αCH), 5.70 (2 H, s, NH₂), 6.23 (1 H, d, J8.0, NH) and 7.19-7.45 (5 H, m, Ar–H); δ_c(50.31 MHz; [²H₆]DMSO) 37.79 (CH₂), 53.92 (aCH), 126.65 (Ar-CH para), 128.41 (Ar-CH ortho), 129.47 (Ar-CH meta), 137.70 (Ar-C quaternary), 158.53 (CONH₂) and 174.17 (CO₂H); *m*/*z* (EI) 208 (6%, M⁺), 148 (86, $[M - NHCONH_2 + H]^+)$, 120 (41, $C_6H_5CH_2CHNH^+)$, 91 $(100, C_7H_7^+), 74 (61, NHCHCO_2H^+) and 65 (28, C_5H_5^+).$

(2S)-N-Aminophenylalanine 33

To an ice-cooled solution of (2.5)-2-benzylhydantoic acid **32** (1.00 g, 4.8 mmol) in aqueous potassium hydroxide (2.5 mol dm⁻³; 7 cm³) was added aqueous potassium hypochlorite (1.37 mol dm⁻³; 4.4 cm³). After 5 min at 0 °C, the solution was heated at 80 °C for 1.5 h, after which toluene (20 cm³), hydrazine hydrate (0.3 cm³) and concentrated hydrochloric acid (35%, 4 cm³) were added consecutively.

Heating was continued for a further 30 min, the mixture was cooled to room temperature, the phases separated, and the solvent removed from the aqueous phase under reduced pressure to yield a mixture of salts. The salts were extracted with hot ethanol (3×10 cm³) and the extracts brought to pH 6.2 with diethylamine. The white solid thus formed was filtered off, washed with ethanol and dried under vacuum to yield (2.S)-N-aminophenylalanine **33** (0.31 g, 36%), mp 199–201 °C (HRMS: found M⁺, 180.0904. Calc. for C₉H₁₂N₂O₂: 180.0899); $[a]_{D}^{22} = 8.0$

(c 1.0 in 5 mol dm⁻³ aq. HCl) [lit.,²³ – 15.8 (c 0.5 in 6 mol dm⁻³ aq. HCl)]; ν_{max} (Nujol)/cm⁻¹ 2900 (OH), 1735 (CO), 1377 (C–O) and 1075 (C–N); $\delta_{\rm H}$ (200 MHz; ²H₂O) 2.96–3.16 (2 H, m, CH₂), 3.77 (1 H, t, *J* 6.2, α CH) and 7.16–7.42 (5 H, m, ArH); $\delta_{\rm C}$ (74.76 MHz; ²H₂O) 33.49 (CH₂), 63.46 (α CH), 125.54 (Ar–CH *para*), 127.03 (Ar–CH *ortho*), 127.36 (Ar–CH *meta*), 133.67 (Ar–C quaternary) and 172.76 (CO₂H); *m*/*z* (EI) 180 (5%, M⁺), 91 (100, C₇H₇⁺), 89 (100, [M – C₇H₇]⁺), 77 (23, C₆H₅⁺), 71 (91, [COCH(CH₂)NH + H]⁺), 65 (35, C₅H₅⁺) and 43 (46, [CH₂-CHNH + H]⁺).

(2S)-N-(Benzyloxycarbonylamino)phenylalanine 34

To an ice-cooled solution of (2S)-N-aminophenylalanine 33 (1.50 g, 8.3 mmol) in aqueous sodium hydroxide $(0.2 \text{ mol dm}^{-3},$ 40 cm³) was added benzyl chloroformate (1.42 g, 8.4 mmol) and the mixture was allowed to reach room temperature whilst stirring vigorously over 1 h. Additional aqueous sodium hydroxide $(1 \text{ mol} \text{dm}^{-3}; 10 \text{ cm}^3)$ was then added to redissolve the white precipitate. The solution was washed with diethyl ether (2 \times 30 cm³) and the aqueous phase acidified with dilute hydrochloric acid (0.5 mol dm⁻³) to pH 3. The resulting precipitate was filtered off, washed with cold water and dried under reduced pressure to give the hydrazide 34 as a white solid (2.06 g, 79%), mp 163-165 °C (Found: C, 64.7; H, 5.8; N, 8.9. C₁₇H₁₈N₂O₄ requires C, 65.0; H, 5.8; N, 8.9%); [a]_D²² -3.9 (c 0.5 in MeOH); v_{max}(Nujol)/cm⁻¹ 3365 (NH), 3287 (NH), 2900 (OH), 1749 (acid CO), 1710 (amide CO) and 1273 (C–O); $\delta_{\rm H}$ (200 MHz; [²H₄]methanol) 2.96–3.01 (2 H, m, CH₂Ph), 3.85 (1 H, m, αCH), 5.10 (2 H, s, OCH₂Ph) and 7.21–7.34 (10 H, m, Ar–H); $\delta_{\rm C}$ (74.76 MHz; [²H₄]methanol) 38.07 (CH₂C₆H₅), 65.92 (αCH), 68.12 (OCH₂C₆H₅), 127.99, 129.23, 129.40, 129.78, 130.67 (Ar-CH), 138.26, 138.71 (Ar-C quaternary), 159.51 (CO2CH2Ph) and 176.13 (CO₂H); m/z (EI) 314 (13%, M⁺), 223 (42, $[M - CH_2C_6H_5]^+)$, 179 (22, $[M - CO_2CH_2C_6H_5]^+)$, 91 (100, $C_7H_7^+$), 77 (22, $C_6H_5^+$) and 65 (34, $C_5H_5^+$).

Methyl (2S)-N-(benzyloxycarbonylamino)phenylalaninate 35

To ice-cooled methanol (5 cm³) was added, dropwise, thionyl chloride (0.14 cm³, 1.9 mmol), followed by hydrazide **34** (0.50 g, 1.6 mmol). The resulting solution was refluxed for 40 min, and allowed to cool to room temperature. The solvent and excess thionyl chloride were removed under reduced pressure and the resulting pale yellow oil was redissolved in ethyl acetate (40 cm³). The solution was washed with 5% aqueous sodium hydrogen carbonate (2×20 cm³), water (2×10 cm³) and then dried (MgSO₄). The solvent was removed under reduced pressure to give a pale yellow solid which was recrystallised from ethyl acetate-light petroleum to give colourless crystals (0.39 g, 74%), mp 62-64 °C (Found: C, 65.5; H, 6.0; N, 8.5. C₁₈H₂₀N₂O₄ requires C, 65.8; H, 6.1; N, 8.5%) (HRMS: found M⁺, 328.1414. $C_{18}H_{20}N_2O_4$ requires 328.1423); $[a]_D^{22}$ +10.9 (c 0.2 in MeOH); $v_{\rm max}$ (Nujol)/cm⁻¹ 1735 (CO) and 1406 (ester C–O); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 2.93-3.16 (2 H, m, CH₂Ph), 3.70 (3 H, s, CO₂CH₃), 4.00 (1 H, t, J 6.3, aCH), 4.20 (1 H, br s, NH), 5.09 (2 H, s, OCH₂Ph), 6.52 (1 H, br s, NH) and 7.22-7.31 (10 H, m, Ar–H); δ_c(50.31 MHz; C²HCl₃), 37.74 (CH₂Ph), 52.48 (CO_2CH_3) , 64.43 (α CH), 67.50 ($OCH_2C_6H_5$), 127.45, 128.66, 128.71, 128.99, 129.06, 129.61 (Ar-CH), 136.55, 136.83 (Ar-C quaternary), 157.42 (CO2CH2C6H5) and 173.57 (CO2CH3); m/z (EI) 328 (5%, M⁺), 237 (57, [M - CH₂C₆H₅]⁺), 193 (35, $[M - CO_2CH_2C_6H_5]^+$), 162 (26, $H_3CO_2CCCH_2C_6H_5^+$), 105 (17, $[CHCH_2C_6H_5 + H]^+$), 91 (100, $C_7H_7^+$) and 65 (22, $C_5H_5^+$).

Methyl (2.5)-*N*-(benzyloxycarbonylamino)-*N*-[(2.5)-*N*-chloroacetylprolyl]phenylalaninate 36

To a stirred solution of (2.S)-*N*-chloroacetylproline **19** (0.53 g, 2.8 mmol) and pyridine (0.31 cm³, 3.9 mmol) in CH₂Cl₂ (10 cm³) was added, dropwise, thionyl chloride (0.26 cm³, 3.6 mmol). After 10 min, a solution of the hydrazide **35** (0.60 g, 1.8 mmol) and DMAP (0.49 g, 4.0 mmol) in CH₂Cl₂ (10 cm³) were added

in one portion. After 3-4 days, the dark brown solution was washed with 0.5 mol dm⁻³ hydrochloric acid (2×20 cm³), 5% aqueous sodium hydrogen carbonate $(2 \times 20 \text{ cm}^3)$ and brine (30 cm³). The organic phase was dried (MgSO₄), and the solvent removed under reduced pressure to give a brown oil which was purified by silica chromatography using light petroleum-ethyl acetate (3:2) as the eluent to give the desired product 36 as a light oil (0.49 g, 53%) (HRMS: found M⁺, 501.1675. C₂₅H₂₈ClN₃O₆ requires 501.1667); $[a]_{D}^{22}$ -66.8 (c 1.1 in MeOH); v_{max} (thin film)/ cm⁻¹ 3241 (NH), 1744 (ester CO), 1690 (secondary amide CO) and 1645 (tertiary amide CO); $\delta_{\rm H}$ (200 MHz; C²HCl₃) 1.92–2.06 (3 H, m, ¹/₂BCH₂ and γCH₂), 2.11-2.18 (1 H, m, ¹/₂BCH₂), 3.00-3.21 (2 H, m, CH₂C₆H₅), 3.55-3.65 (2 H, m, \deltaCH₂), 3.70 (3 H, s, CO₂CH₃), 3.93-4.08 (2 H, dd, J₁ 13.2, J₂ 4.0, CH₂Cl), 4.61 [t, 1 H, m, αCH(Pro)], 4.73 [c, 1 H, m, αCH(Pro)], 5.10-5.25 [3 H, m; c, αCH(Phe); and CO₂CH₂Ph], 5.50 [t, 1 H, m, αCH(Phe)], 7.25–7.39 (5 H, m, Ar–H) and 8.10 (1 H, br s, NH); $\delta_{\rm C}$ (74.76 MHz; C²HCl₃) 22.40 (c, γ CH₂), 25.38 (t, γ CH₂), 29.11 (t, β CH₂), 31.00 (c, β CH₂), 34.79 (CH₂C₆H₅), 42.09 (c, CH₂Cl), 42.16 (t, CH₂Cl), 47.74 (\deltaCH₂), 52.51 (c, CO₂CH₃), 53.02 (t, CO₂CH₃), 60.42 (αCH), 68.21 (c, OCH₂Ph), 68.80 (t, OCH₂Ph), 127.10-129.63 (Ar-CH), 136.15, 137.09 (Ar-C quaternary), 155.94 (CO2CH2Ph), 165.62 (COCH2Cl), 169.93 (CON) and 174.37 (CO₂CH₃); m/z (EI) 502 (9%, M⁺), 328 (31, [M - $ClCH_{2}CONC_{4}H_{7}CO + H]^{+}$), 237 (21, [M - ClCH₂CONC₄H₇- $CO - C_7H_7 + H]^+$), 174 (57, $ClCH_2CONC_4H_7CO^+$), 146 $(100, ClCH_2CONC_4H_7^+)$, 91 (71, $C_7H_7^+$) and 70 (74, $[C_4H_7N +$ H]⁺).

(9a.5)-2-[(1.5)-1-Methoxycarbonyl-2-phenylethyl]-2,3,4,5,7,8,9, 9a-octahydro-1*H*-pyrrolo[2,1-*d*][1,2,5]triazepine-1,5-dione 29

To the dipeptide 36 (100 mg, 0.2 mmol) was added a solution of hydrogen bromide in acetic acid (30%; 6 cm³). The solution was stirred at room temperature for 2 h and then poured into diethyl ether (100 cm³), resulting in the formation of a flocculent white precipitate (this decomposed to a brown oil if exposed to air). The solvents were carefully decanted, the suspension diluted with diethyl ether (50 cm³), shaken, and the solvent again decanted. The remaining solvents were removed under reduced pressure and the residue redissolved in aqueous sodium hydroxide (0.01 mol dm⁻³; 20 cm³) to give neutral pH. The solution was allowed to stand for 90 min, and extracted with ethyl acetate $(3 \times 20 \text{ cm}^3)$. The organic phase was washed with water (50 cm³), dried (MgSO₄) and the solvent removed under reduced pressure to give a clear oil which crystallised from ethyl acetatelight petroleum to give pale yellow crystals (60 mg, 91%), mp 114-116 °C (Found: C, 49.5; H, 5.5; N, 10.0. C₁₇H₂₁N₃O₄·HBr requires C, 49.5; H, 5.4; N, 10.2%) (HRMS: found M⁺, 331.1523. $C_{17}H_{21}N_3O_4$ requires 331.1532); $[a]_D^{22} - 78.3$ (c 0.8 in MeOH); v_{max}(thin film)/cm⁻¹ 3353 (NH), 1740 (ester CO), 1672 (amide CO, br), 1170 (C–O) and 1093 (C–N); $\delta_{\rm H}$ (300 MHz; $C^{2}HCl_{3}$) 1.75–2.15 (3 H, m, $\frac{1}{2}\beta CH_{2}$ and γCH_{2}), 2.18 (A, 1 H, m, ½βCH₂), 2.32 (B, 1 H, m, ½βCH₂), 2.61 (1 H, d, J 11.3, ¹/₂COCH₂N), 2.99-3.50 (2 H, m, CH₂C₆H₅), 3.09 (1 H, d, J11.3, ¹/₂COCH₂N), 3.50-3.82 (2 H, m, δCH₂), 3.67 (A or B, 3 H, s, CO₂CH₃), 3.76 (A or B, 3 H, s, CO₂CH₃), 5.12 [B, 1 H, dd, J₁ 8.9, J₂ 3.0, αCH(Pro)], 5.29 [A, 1 H, dd, J₁ 8.9, J₂ 3.0, αCH(Pro)], 5.45 [A or B, 1 H, dd, J₁ 9.2, J₂ 6.3, αCH(Phe)], 5.65 [A or B, 1 H, dd J₁ 12.3, J₂ 4.8, αCH(Phe)] and 7.19-7.33 (5 H, m, Ar-H); δ_C(74.56 MHz; C²HCl₃) 22.36, 24.66 (γCH₂), 27.29, 27.42 (COCH₂N), 29.00, 31.52 (βCH₂), 33.59, 34.10 (CH₂C₆H₅), 47.80, 48.12 (δCH₂), 52.41, 52.77 (CO₂CH₃), 57.57, 58.13 [aCH(Pro)], 57.96, 58.69 [aCH(Phe)], 127.27, 127.81 (Ar-CH para), 128.39, 128.82 (Ar-CH ortho), 129.05, 129.36 (Ar-CH meta), 135.90, 136.42 (Ar-C quaternary), 165.14, 165.25 (COCH₂N), 171.42, 172.05 (CON) and 174.71 (CO₂CH₃); m/z (EI) 331 (M⁺, 53%), 212 (63, [M - CO- $CH_2C_6H_5]^+$), 168 (38, $[M - C_6H_5CH_2CHCO_2CH_3]^+$), 125 (39, $NC_4H_7CONCH^+$), 112 (51, $NC_4H_9CON^+$), 91 (48, $C_7H_7^+$) and 70 (100, C₄H₈⁺).

(9a.S)-2-[(1.S)-1-Methoxycarbonyl-2-phenylethyl]-3-[(2.S)-*N*benzyloxycarbonylalanyl]-2,3,4,5,7,8,9,9a-octahydro-1*H*pyrrolo[2,1-*d*][1,2,5]triazepine-1,5-dione 30

To a solution of *N*-methylmorpholine (40 mm³, 0.36 mmol) in dry THF (3 cm³) was added (2.S)-N-benzyloxycarbonylalanine (76 mg, 0.34 mmol) and the solution cooled to -15 °C. Isobutyl chloroformate (50 mm³, 0.32 mmol) was added with stirring and the resulting suspension was stirred at -15 °C for 2 min. A mixture of triazepine-1,5-dione 29 (100 mg, 0.30 mmol) and Nmethylmorpholine (40 mm³, 0.36 mmol) in dry DMF (1 cm³) was then added in one portion to the cold suspension. The reaction mixture was allowed to warm to room temperature and stirred for a further 5 h. The hydrochloride salts were filtered off and the solvents were removed under reduced pressure. The resulting oil was dissolved in ethyl acetate (10 cm³) washed with 0.5 mol dm⁻³ aqueous HCl (2×5 cm³), 5% aqueous sodium hydrogen carbonate $(2 \times 5 \text{ cm}^3)$ and brine (5 cm³), dried (MgSO₄) and the solvent removed under reduced pressure to yield the product as a pale yellow oil. Purification by silica column chromatography using ethyl acetate as the eluent gave the pure product as a clear oil (6 mg, 4%) (HRMS: found M^+ , 536.2283. $C_{28}H_{32}N_4O_7$ requires 536.2271); v_{max} (thin film)/ cm⁻¹ 1742 (ester CO), 1725 (urethane CO), 1680 (secondary amide CO), 1650 (tertiary amide CO) and 1180 (C–O); $\delta_{\rm H}$ (300 MHz; C²HCl₃) 1.42 (3 H, d, J7.1, CHCH₃), 1.79-2.27 (4 H, m, γCH₂ and βCH₂), 3.18-3.49 (2 H, m, CH₂Ph), 3.62-3.87 (2 H, m, δCH_2), 3.71 (A, 3 H, s, OCH₃), 3.76 (B, 3 H, s, OCH₃), 4.00-4.08 (2 H, m, CH₂N), 4.36 [1 H, m, αCH (Ala)], 5.11 (2 H, s, OCH₂Ph), 5.21-5.42 [2 H, αCH (Pro and Phe)], 6.64 (A or B, 1 H, m, NH), 6.80 (A or B, 1 H, m, NH) and 7.07-7.36 (10 H, m, Ar-H); δ_c(74.76 MHz; C²HCl₃) 18.65 (CH₃), 22.27, 24.99 (γCH₂), 29.03, 31.61 (βCH₂), 33.39, 34.33 (CH₂C₆H₅), 41.79, 42.01 (CH₂N), 47.67, 47.97 (δCH₂), 49.70 [αCH(Ala)], 52.68, 52.97 (CO_2CH_3) , 58.43, 58.69 $[\alpha CH(Pro)]$, 64.10, 64.33 [aCH(Phe)], 67.05 (OCH2Ph), 127.04, 127.48, 128.24, 128.36, 128.70, 128.75, 128.89, 129.16 and 129.39 (Ar-CH), 136.65 and 137.76 (Ar-CH₂ quaternary), 141.71 and 143.09 (Ar-CH₂O quaternary), 155.83 (CO2CH2Ph), 164.95, 169.79 and 172.33 (CO); m/z (EI) 536 (M⁺, 8%), 190 [27, OCNCH(CO₂)-CH₂C₆H₅⁺], 174 [30, OCNCH(CO)CH₂C₆H₅⁺], 146 (66, OCNCHCH₂C₆H₅⁺), 140 (29, CH₃CONC₄H₇CO⁺), 112 (45, $NC_4H_8CON^+$), 91 (100, $C_7H_7^+$) and 70 (96, $C_4H_8N^+$).

Acknowledgements

We thank the BBSRC for research grant 49/P01078, Dr T. Jenn for the synthesis of compound **16**, the University of St Andrews for a studentship to M. M. L., The Wellcome Trust for a research Studentship to A. L. and Dr M. Akhtar for scientific support.

References

- 1 Please note that in the preliminary communication on this work (M. M. Lenman, S. L. Ingham and D. Gani, *Chem. Commun.*, 1996, 85), we referred to the fused system **6** as a 1,2,5-triazepine-3,6-dione and **12** as a 2,5-diketopiperazine. The numbering of the diones was based upon the unfused compounds, we apologise for any confusion this may have caused.
- 2 W. S. Blair and B. L. Semler, *Curr. Opin. Cell Biol.*, 1991, **3**, 1039, and references cited therein.
- 3 P. N. Lewis, F. A. Momany and H. A. Scheraga, *Biochim. Biophys. Acta*, 1973, **303**, 211.
- 4 R. E. London, J. M. Stewart, R. Williams, J. R. Cann and N. A. Matwiyoff, *J. Am. Chem. Soc.*, 1979, **101**, 2455.
- 5 R. L. Stein, Adv. Protein Chem., 1993, 44, 1.
- 6 G. Fischer, B. Wittmann-Liebold, K. Lang, T. Kiefhaber and F. X. Schmid, *Nature*, 1989, **337**, 476.
- 7 M. W. Harding, A. Galat, D. E. Uehling and S. L. Schreiber, *Nature*, 1989, **341**, 756.
- 8 G. Fischer, Angew. Chem., Int. Ed. Engl., 1994, 33, 1415, and references cited therein.

2310 J. Chem. Soc., Perkin Trans. 1, 1997

- 9 M. Liakopoulou-Kyriakides and R. E. Galardy, *Biochemistry*, 1979, 18, 1952.
- 10 D. Gramberg and J. A. Robinson, *Tetrahedron Lett.*, 1994, **35**, 861.
- T. P. Curran and P. M. McEnaney, *Tetrahedron Lett.*, 1995, **36**, 191.
 G. D. Rose, L. M. Gierasch and J. A. Smith, *Adv. Protein Chem.*,
- 1985, **37**, 1, and references cited therein. 13 J. S. Nowick, E. M. Smith and M. Pairish, *Chem. Soc. Rev.*, 1996,
- 401, and references therein.
 14 N. De la Figuera, I. Alkorta, M. T. García-López, R. Herranz and R. González-Muñiz, *Tetrahedron Lett.*, 1995, **51**, 7841.
- 15 D. Obercht, U. Bohdal, J. Daly, C. Lehmann, P. Schönholzer and K. Müller, *Tetrahedron*, 1995, **51**, 10883.
- 16 D. C. Horwell, D. Taylor and H. M. G. Williams, *Bioorg. Med. Chem. Lett.*, 1997, 7, 31.
- 17 B. Hartzoulakis, T. J. Rutherford, M. D. Ryan and D. Gani, *Tetrahedron Lett.*, 1996, 37, 6911.
- 18 D. S. Kemp and Z. Q. Li, *Tetrahedron Lett.*, 1995, **36**, 4175.
- 19 J. Vicar, J. Smolíková and K. Bláha, *Collect. Czech. Chem. Commun.*, 1972, **37**, 4060.

- 20 S.-C. J. Fu, S. M. Birnbaum and J. P. Greenstein, J. Am. Chem. Soc., 1954, 76, 6054.
- 21 M. D. Ryan and J. Drew, EMBO J., 1994, 13, 928.
- M. L. L. Donnelly, D. Gani, M. Flint, S. Monagham and M. Ryan, J. Gen. Virol., 1997, 78, 13.
- 23 J. Viret, J. Gabard and A. Collet, Tetrahedron, 1987, 43, 891.
- 24 W. C. Still, M. Khan and A. Mitra, J. Org. Chem., 1978, 43, 2924.
- 25 St. Guttmann, *Helv. Chim. Acta*, 1961, **44**, 721.
- 26 K. Freudenberg and L. Markert, *Chem. Ber.*, 1927, **60**, 2447.
 27 I. Z. Siemion, *Org. Magn. Reson.*, 1971, **3**, 545.
- 28 E. Ronwin, *J. Org. Chem.*, 1953, **18**, 127.
- 29 C. T. Pedersen, *Acta Chem. Scand.*, 1964, **18**, 2199.

Paper 7/02107K Received 26th March 1997 Accepted 14th April 1997